

# PERIODIC TABLE Atomic Properties of the Elements

**Frequently used fundamental physical constants**  
For the most accurate values of these and other constants, visit [physics.nist.gov/constants](http://physics.nist.gov/constants)  
1 second = 9 192 631 770 periods of radiation corresponding to the transition between the two hyperfine levels of the ground state of <sup>133</sup>Cs

speed of light in vacuum	<i>c</i>	299 792 458 m s <sup>-1</sup>	(exact)
Planck constant	<i>h</i>	6.6261 × 10 <sup>-34</sup> J s	( <i>h</i> = <i>h</i> /2π)
elementary charge	<i>e</i>	1.6022 × 10 <sup>-19</sup> C	
electron mass	<i>m<sub>e</sub></i>	9.1094 × 10 <sup>-31</sup> kg	
	<i>m<sub>e</sub>c<sup>2</sup></i>	0.5110 MeV	
proton mass	<i>m<sub>p</sub></i>	1.6726 × 10 <sup>-27</sup> kg	
fine-structure constant	<i>α</i>	1/137.036	
Rydberg constant	<i>R<sub>∞</sub></i>	10 973 732 m <sup>-1</sup>	
	<i>R<sub>∞</sub>c</i>	3.289 842 × 10 <sup>15</sup> Hz	
	<i>R<sub>∞</sub>hc</i>	13.6057 eV	
Boltzmann constant	<i>k</i>	1.3807 × 10 <sup>-23</sup> J K <sup>-1</sup>	

- Solids
- Liquids
- Gases
- Artificially Prepared

13 IIIA		14 IVA		15 VA		16 VIA		17 VIIA		18 VIIIa	
5 <b>B</b> Boron 10.811 1s <sup>2</sup> 2s <sup>2</sup> 2p 8.2980	6 <b>C</b> Carbon 12.0107 1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>2</sup> 11.2603	7 <b>N</b> Nitrogen 14.0067 1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>3</sup> 14.5341	8 <b>O</b> Oxygen 15.9994 1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>4</sup> 13.6181	9 <b>F</b> Fluorine 18.9984032 1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>5</sup> 17.4228	10 <b>Ne</b> Neon 20.1797 1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup> 21.5645	13 <b>Al</b> Aluminum 26.981538 [Ne]3s <sup>2</sup> 3p 5.9858	14 <b>Si</b> Silicon 28.0855 [Ne]3s <sup>2</sup> 3p <sup>2</sup> 8.1517	15 <b>P</b> Phosphorus 30.973761 [Ne]3s <sup>2</sup> 3p <sup>3</sup> 10.4867	16 <b>S</b> Sulfur 32.065 [Ne]3s <sup>2</sup> 3p <sup>4</sup> 10.3600	17 <b>Cl</b> Chlorine 35.453 [Ne]3s <sup>2</sup> 3p <sup>5</sup> 12.9676	18 <b>Ar</b> Argon 39.948 [Ne]3s <sup>2</sup> 3p <sup>6</sup> 15.7596

1 <b>H</b> Hydrogen 1.00794 1s 13.5984	2 <b>He</b> Helium 4.002602 1s <sup>2</sup> 24.5874
3 <b>Li</b> Lithium 6.941 1s <sup>2</sup> 2s 5.3917	4 <b>Be</b> Beryllium 9.012182 1s <sup>2</sup> 2s <sup>2</sup> 9.3227
11 <b>Na</b> Sodium 22.989770 [Ne]3s 5.1391	12 <b>Mg</b> Magnesium 24.3050 [Ne]3s <sup>2</sup> 7.6462
19 <b>K</b> Potassium 39.0983 [Ar]4s 4.3407	20 <b>Ca</b> Calcium 40.078 [Ar]4s <sup>2</sup> 6.1132
37 <b>Rb</b> Rubidium 85.4678 [Kr]5s 4.1771	38 <b>Sr</b> Strontium 87.62 [Kr]5s <sup>2</sup> 5.6949
55 <b>Cs</b> Cesium 132.90545 [Xe]6s 3.8939	56 <b>Ba</b> Barium 137.327 [Xe]6s <sup>2</sup> 5.2117
87 <b>Fr</b> Francium (223) [Rn]7s 4.0727	88 <b>Ra</b> Radium (226) [Rn]7s <sup>2</sup> 5.2784

21 <b>Sc</b> Scandium 44.955910 [Ar]3d4s <sup>2</sup> 6.5615	22 <b>Ti</b> Titanium 47.867 [Ar]3d <sup>2</sup> 4s <sup>2</sup> 6.8281	23 <b>V</b> Vanadium 50.9415 [Ar]3d <sup>3</sup> 4s <sup>2</sup> 6.7462	24 <b>Cr</b> Chromium 51.9961 [Ar]3d <sup>5</sup> 4s 6.7665	25 <b>Mn</b> Manganese 54.938049 [Ar]3d <sup>5</sup> 4s <sup>2</sup> 7.4340	26 <b>Fe</b> Iron 55.845 [Ar]3d <sup>6</sup> 4s <sup>2</sup> 7.9024	27 <b>Co</b> Cobalt 58.933200 [Ar]3d <sup>7</sup> 4s <sup>2</sup> 7.8810	28 <b>Ni</b> Nickel 58.6934 [Ar]3d <sup>8</sup> 4s <sup>2</sup> 7.6398	29 <b>Cu</b> Copper 63.546 [Ar]3d <sup>10</sup> 4s 7.7264	30 <b>Zn</b> Zinc 65.409 [Ar]3d <sup>10</sup> 4s <sup>2</sup> 9.3942	31 <b>Ga</b> Gallium 69.723 [Ar]3d <sup>10</sup> 4s <sup>2</sup> 4p 5.9993	32 <b>Ge</b> Germanium 72.64 [Ar]3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>2</sup> 7.8994	33 <b>As</b> Arsenic 74.92160 [Ar]3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>3</sup> 9.7886	34 <b>Se</b> Selenium 78.96 [Ar]3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>4</sup> 9.7524	35 <b>Br</b> Bromine 79.904 [Ar]3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>5</sup> 11.8138	36 <b>Kr</b> Krypton 83.798 [Ar]3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>6</sup> 13.9996
39 <b>Y</b> Yttrium 88.90585 [Kr]4d5s <sup>2</sup> 6.2173	40 <b>Zr</b> Zirconium 91.224 [Kr]4d <sup>5</sup> 5s <sup>2</sup> 6.6339	41 <b>Nb</b> Niobium 92.90638 [Kr]4d <sup>5</sup> 5s 6.7589	42 <b>Mo</b> Molybdenum 95.94 [Kr]4d <sup>5</sup> 5s 7.0924	43 <b>Tc</b> Technetium (98) [Kr]4d <sup>5</sup> 5s <sup>2</sup> 7.28	44 <b>Ru</b> Ruthenium 101.07 [Kr]4d <sup>7</sup> 5s 7.3605	45 <b>Rh</b> Rhodium 102.90550 [Kr]4d <sup>8</sup> 5s 7.4589	46 <b>Pd</b> Palladium 106.42 [Kr]4d <sup>10</sup> 5s 8.3369	47 <b>Ag</b> Silver 107.8682 [Kr]4d <sup>10</sup> 5s 7.5762	48 <b>Cd</b> Cadmium 112.411 [Kr]4d <sup>10</sup> 5s <sup>2</sup> 8.9938	49 <b>In</b> Indium 114.818 [Kr]4d <sup>10</sup> 5s <sup>2</sup> 5p 5.7864	50 <b>Sn</b> Tin 118.710 [Kr]4d <sup>10</sup> 5s <sup>2</sup> 5p <sup>2</sup> 7.3439	51 <b>Sb</b> Antimony 121.760 [Kr]4d <sup>10</sup> 5s <sup>2</sup> 5p <sup>3</sup> 8.6084	52 <b>Te</b> Tellurium 127.60 [Kr]4d <sup>10</sup> 5s <sup>2</sup> 5p <sup>4</sup> 9.0096	53 <b>I</b> Iodine 126.90447 [Kr]4d <sup>10</sup> 5s <sup>2</sup> 5p <sup>5</sup> 10.4513	54 <b>Xe</b> Xenon 131.293 [Kr]4d <sup>10</sup> 5s <sup>2</sup> 5p <sup>6</sup> 12.1298
72 <b>Hf</b> Hafnium 178.49 [Xe]4f <sup>14</sup> 5d <sup>4</sup> 6s <sup>2</sup> 6.8251	73 <b>Ta</b> Tantalum 180.9479 [Xe]4f <sup>14</sup> 5d <sup>5</sup> 6s <sup>2</sup> 7.5496	74 <b>W</b> Tungsten 183.84 [Xe]4f <sup>14</sup> 5d <sup>6</sup> 6s <sup>2</sup> 7.8640	75 <b>Re</b> Rhenium 186.207 [Xe]4f <sup>14</sup> 5d <sup>7</sup> 6s <sup>2</sup> 7.8335	76 <b>Os</b> Osmium 190.23 [Xe]4f <sup>14</sup> 5d <sup>8</sup> 6s <sup>2</sup> 8.4382	77 <b>Ir</b> Iridium 192.217 [Xe]4f <sup>14</sup> 5d <sup>9</sup> 6s <sup>2</sup> 8.9670	78 <b>Pt</b> Platinum 195.078 [Xe]4f <sup>14</sup> 5d <sup>9</sup> 6s <sup>1</sup> 8.9588	79 <b>Au</b> Gold 196.96655 [Xe]4f <sup>14</sup> 5d <sup>10</sup> 6s 9.2255	80 <b>Hg</b> Mercury 200.59 [Xe]4f <sup>14</sup> 5d <sup>10</sup> 6s <sup>2</sup> 10.4375	81 <b>Tl</b> Thallium 204.3833 [Hg]6p 6.1082	82 <b>Pb</b> Lead 207.2 [Hg]6p <sup>2</sup> 7.4167	83 <b>Bi</b> Bismuth 208.98038 [Hg]6p <sup>3</sup> 7.2855	84 <b>Po</b> Polonium (209) [Hg]6p <sup>4</sup> 8.414	85 <b>At</b> Astatine (210) [Hg]6p <sup>5</sup>	86 <b>Rn</b> Radon (222) [Hg]6p <sup>6</sup> 10.7485	
104 <b>Rf</b> Rutherfordium [Rn]5f <sup>14</sup> 6d <sup>2</sup> 7s <sup>2</sup> ? 6.0 ?	105 <b>Db</b> Dubnium (262)	106 <b>Sg</b> Seaborgium (266)	107 <b>Bh</b> Bohrium (264)	108 <b>Hs</b> Hassium (277)	109 <b>Mt</b> Meitnerium (268)	110 <b>Uun</b> Ununnilium (281)	111 <b>Uuu</b> Unununium (272)	112 <b>Uub</b> Ununbium (285)	114 <b>Uuq</b> Ununquadium (289)	116 <b>Uuh</b> Ununhexium (292)					

Atomic Number: 58  
Ground-state Level: 1G<sub>4</sub>  
Symbol: **Ce**  
Name: Cerium  
Atomic Weight: 140.116  
Ground-state Configuration: [Xe]4f5d6s<sup>2</sup>  
Ionization Energy (eV): 5.5387

57 <b>La</b> Lanthanum 138.9055 [Xe]5d6s <sup>2</sup> 5.5769	58 <b>Ce</b> Cerium 140.116 [Xe]4f5d6s <sup>2</sup> 5.5387	59 <b>Pr</b> Praseodymium 140.90765 [Xe]4f <sup>3</sup> 6s <sup>2</sup> 5.473	60 <b>Nd</b> Neodymium 144.24 [Xe]4f <sup>4</sup> 6s <sup>2</sup> 5.5250	61 <b>Pm</b> Promethium (145) [Xe]4f <sup>5</sup> 6s <sup>2</sup> 5.582	62 <b>Sm</b> Samarium 150.36 [Xe]4f <sup>6</sup> 6s <sup>2</sup> 5.6437	63 <b>Eu</b> Europium 151.964 [Xe]4f <sup>7</sup> 6s <sup>2</sup> 5.6704	64 <b>Gd</b> Gadolinium 157.25 [Xe]4f <sup>7</sup> 5d6s <sup>2</sup> 6.1498	65 <b>Tb</b> Terbium 158.92534 [Xe]4f <sup>9</sup> 6s <sup>2</sup> 5.8638	66 <b>Dy</b> Dysprosium 162.500 [Xe]4f <sup>10</sup> 6s <sup>2</sup> 5.9389	67 <b>Ho</b> Holmium 164.93032 [Xe]4f <sup>11</sup> 6s <sup>2</sup> 6.0215	68 <b>Er</b> Erbium 167.259 [Xe]4f <sup>12</sup> 6s <sup>2</sup> 6.1077	69 <b>Tm</b> Thulium 168.93421 [Xe]4f <sup>13</sup> 6s <sup>2</sup> 6.1843	70 <b>Yb</b> Ytterbium 173.04 [Xe]4f <sup>14</sup> 6s <sup>2</sup> 6.2542	71 <b>Lu</b> Lutetium 174.967 [Xe]4f <sup>14</sup> 5d6s <sup>2</sup> 5.4259
89 <b>Ac</b> Actinium (227) [Rn]6d7s <sup>2</sup> 5.17	90 <b>Th</b> Thorium 232.0381 [Rn]6d <sup>2</sup> 7s <sup>2</sup> 6.3067	91 <b>Pa</b> Protactinium 231.03588 [Rn]5f <sup>2</sup> 6d7s <sup>2</sup> 5.89	92 <b>U</b> Uranium 238.02891 [Rn]5f <sup>3</sup> 6d7s <sup>2</sup> 6.1941	93 <b>Np</b> Neptunium (237) [Rn]5f <sup>4</sup> 6d7s <sup>2</sup> 6.2657	94 <b>Pu</b> Plutonium (244) [Rn]5f <sup>6</sup> 7s <sup>2</sup> 6.0260	95 <b>Am</b> Americium (243) [Rn]5f <sup>7</sup> 7s <sup>2</sup> 5.9738	96 <b>Cm</b> Curium (247) [Rn]5f <sup>8</sup> 6d7s <sup>2</sup> 5.9914	97 <b>Bk</b> Berkelium (247) [Rn]5f <sup>9</sup> 7s <sup>2</sup> 6.1979	98 <b>Cf</b> Californium (251) [Rn]5f <sup>10</sup> 7s <sup>2</sup> 6.2817	99 <b>Es</b> Einsteinium (252) [Rn]5f <sup>11</sup> 7s <sup>2</sup> 6.42	100 <b>Fm</b> Fermium (257) [Rn]5f <sup>12</sup> 7s <sup>2</sup> 6.50	101 <b>Md</b> Mendelevium (258) [Rn]5f <sup>13</sup> 7s <sup>2</sup> 6.58	102 <b>No</b> Nobelium (259) [Rn]5f <sup>14</sup> 7s <sup>2</sup> 6.65	103 <b>Lr</b> Lawrencium (262) [Rn]5f <sup>14</sup> 7s <sup>2</sup> ? 4.9 ?

<sup>†</sup>Based upon <sup>12</sup>C. ( ) indicates the mass number of the most stable isotope.