

METRIC

MIL-STD-1443

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MILITARY STANDARD

INORGANIC ACIDS AND ACID ANHYDRIDES, REAGENT GRADE

(INCLUDING ACS AND USP-NF COMPOUNDS) (METRIC)



AMSC N/A

FSC 6810

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MIL-STD-1443

DEPARTMENT OF DEFENSE
Washington, DC 20301

Inorganic Acids and Acid Anhydrides, Reagent Grade (Including ACS and USP-NF Compounds)

MIL-STD-1443

1. This Military Standard is approved for use by all Departments and Agencies of the Department of Defense.
2. Beneficial comments (recommendations, additions, deletions) and any pertinent data which may be of use in improving this document should be addressed to: Commander, U.S. Army Chemical Research, Development and Engineering Center, Attn: SMCCR-SPD-TS, Aberdeen Proving Ground, MD 21010-5423, by using the self-addressed Standardization Document Improvement Proposal (DD Form 1426) appearing at the end of this document or by letter.

MIL-STD-1443

FOREWORD

This standard is approved for use by all Departments and Agencies of the Department of Defense in the selection of items for application. It is intended to prevent the entry of unnecessary items (sizes, types, varieties) into the Department of Defense logistics system. This document is not intended to restrict any service in selecting new items resulting from state-of-the-art changes.

MIL-STD-1443

CONTENTS

<u>Paragraph</u>		<u>Page</u>
1.	SCOPE	1
1.1	Coverage	1
1.2	Application	1
1.3	Classification	1
2.	REFERENCED DOCUMENTS	2
2.1	Government documents	2
2.1.1	Specifications, standards, and handbooks	2
2.1.2	Other Government documents, drawings, and publications	2
2.2	Other publications	3
2.3	Order of precedence	3
3.	DEFINITIONS	4
4.	GENERAL REQUIREMENTS	5
4.1	Packaging data and labeling	5
4.2	Hazardous materials information	5
4.3	Safety	5
4.3.1	Personal protective measures	5
4.3.1.1	Respiratory protection	5
4.3.1.2	Skin protection	5
4.3.1.3	Face and eye protection	5
4.3.1.4	Training	5
4.3.1.5	Exercises	6
4.3.2	Storage conditions	6
4.3.2.1	Flammable, combustible, pyrophoric and ignitable materials	6
4.3.2.2	Water-sensitive fire and explosive hazardous materials	6
4.3.2.3	Incompatible materials	6
4.3.3	Chemical hazardous exposure limits	7
4.3.4	Toxicity	7
4.3.4.1	EPA Toxic (T)	7
4.3.4.2	EPA Acute Hazardous Toxicity (H)	7
4.3.4.3	Hazardous toxic constituents	7
4.4	Pollution and disposal	7
4.4.1	Pollution potential	7
4.4.2	Disposal of excess or unserviceable material	7
4.4.3	Disposal and storage of hazardous wastes	8
4.4.3.1	Cleanup of liquid spills	8
4.4.3.2	Ultimate disposal	9
4.4.4	Disclaimer	9
5.	DETAIL REQUIREMENTS	10
5.1	Arsenic Pentoxide, Analyzed Reagent	10
5.2	Arsenic Pentoxide, n-Hydrate Analyzed Reagent.	11
5.3	Arsenic Trioxide, Analyzed Reagent.	12
5.4	Boric Acid, Analyzed Reagent.	13
5.5	Chlorosulfonic Acid, Reagent.	14
5.6	Chromium Trioxide, Analyzed Reagent.	15
5.7	Hydrobromic Acid (48%), Analyzed Reagent.	16
5.8	Hydrochloric Acid, Analyzed Reagent.	17

MIL-STD-1443

CONTENTS - Continued

<u>Paragraph</u>		<u>Page</u>
5.9	Hydrofluoric Acid, Analyzed Reagent.	18
5.10	Iodic Acid, Analyzed Reagent.	19
5.11	Iodic Pentoxide, Analyzed Reagent.. . . .	20
5.12	Molybdenum Trioxide, Analyzed Reagent.. . . .	21
5.13	Nitric Acid, Analyzed Reagent.. . . .	22
5.14	Phosphomolybdic Acid, Analyzed Reagent.	24
5.15	Phosphoric Acid (Meta), Analyzed Reagent.	25
5.16	Phosphoric Acid (Ortho) , Analyzed Reagent.	26
5.17	Phosphorous Acid (Hypo), 50%, Analyzed Reagent.	27
5.18	Phosphorous Acid, Analyzed Reagent.	28
5.19	Phosphorus Pentoxide, Analyzed Reagent.	29
5.20	Selenious Acid, Analyzed Reagent.	30
5.21	Sulfamic Acid, Analyzed Reagent.. . . .	31
5.22	Sulfuric Acid, Analyzed Reagent.. . . .	32
5.23	Sulfuric Acid, 18N, Analyzed Reagent.	33
5.24	Sulfuric Acid, Fuming, Analyzed Reagent.. . . .	35
5.25	Sulfurous Acid, Analyzed Reagent.	36
5.26	Sulfur Trioxide, Stabilized, Analyzed Reagent.. . . .	37
6.	NOTES	39

MIL-STD-1443

TABLES

I.	Arsenic pentoxide, analyzed reagent - requirements.	10
II.	Arsenic pentoxide, n-hydrate analyzed reagent - requirements.	11
III.	Arsenic trioxide, analyzed reagent - requirements (ACS).. .	12
IV.	Boric acid, analyzed reagent - requirements (ACS).. . . .	13
V.	Chlorosulfonic acid, reagent - requirements.. . . .	14
VI.	Chromium trioxide, analyzed reagent - requirements (ACS)..	15
VII.	Hydrobromic acid (48%), analyzed reagent - requirements (ACS).	16
VIII.	Hydrochloric acid, analyzed reagent - requirements (ACS)..	18
IX.	Hydrofluoric acid, analyzed reagent - requirements (ACS)..	19
X.	Iodic acid, analyzed reagent - requirements (ACS).. . . .	20
XI.	Iodic pentoxide, analyzed reagent - requirements (USP).. .	21
XII.	Molybdenum trioxide, analyzed reagent - requirements (ACS)..	22
XIII.	Nitric acid (69%) analyzed reagent - requirements (ACS).. .	23
XIV.	Nitric acid (90%) analyzed reagent - requirements (ACS).. .	23
XV.	Phosphomolybdic acid, analyzed reagent - requirements (ACS)..	24
XVI.	Phosphoric acid (meta), analyzed reagent - requirements (ACS).	25
XVII.	Phosphoric acid (ortho), analyzed reagent - requirements (ACS).	26
XVIII.	Phosphorous acid (hypo), 50%, analyzed reagent - requirements (USP).	27
XIX.	Phosphorous acid, analyzed reagent - requirements.	28
XX.	Phosphorus pentoxide, analyzed reagent - requirements (ACS)..	29
XXI.	Selenious acid, analyzed reagent - requirements (USP).. . .	30
XXII.	Sulfamic acid, analyzed reagent - requirements (USP).. . .	31
XXIII.	Sulfuric acid, analyzed reagent - requirements (ACS).. . . .	32
XXIV.	Sulfuric acid, 18N, analyzed reagent - requirements. . . .	34
XXV.	Sulfuric acid, fuming, analyzed reagent - requirements (ACS).	35
XXVI.	Sulfuric acid, fuming, analyzed reagent - requirements. . .	35
XXVII.	Sulfurous acid, analyzed reagent - requirements (ACS).. .	37
XXVIII.	Sulfur trioxide, stabilized, analyzed reagent - typical analysis.	37

MIL-STD-1443

1. SCOPE

1.1 Coverage. This standard is a presentation of nomenclature, formulas, physical and chemical properties, specification requirements, military and typical commercial uses, safety information, storage information and disposal information for inorganic acids and acid anhydrides, reagent grade (including ACS and USP-NF Compounds). This standard does not include all of the items represented by the title or all those items which are commercially available. It does contain items preferred for use in the selection of inorganic acids and acid anhydrides, reagent grade (including ACS and USP-NF compounds) for application by the Department of Defense.

1.2 Application. Inorganic acids and acid anhydrides, reagent grade (including ACS and USP-NF compounds) are used as reagents in analytical chemistry.

1.3 Classification. The items in this standard are classified on the basis of chemical composition as inorganic acids and acid anhydrides.

MIL-STD-1443

2. REFERENCED DOCUMENTS

2.1 Government documents.

2.1.1 Specifications, standards, and handbooks. Unless otherwise specified, the following specifications, standards, and handbooks of the issue listed in that issue of the Department of Defense Index of Specifications and Standards (DODISS) specified in the solicitation form a part of this standard to the extent specified herein.

SPECIFICATIONS

FEDERAL

PPP-C-2020 - Chemicals, Liquid, Dry and Paste; Packaging Of
O-C-265 - Chemicals Analytical; General Specification For

MILITARY

MIL-P-13981 - Phosphorous Acid, Analyzed Reagent
MIL-C-51130 - Chemicals, Reagent Grade, General Specification For
MIL-A-51313 - Arsenic Pentoxide, Analyzed Reagent

STANDARDS

FEDERAL

FED-STD-313 - Material Safety Data Sheets, Preparation And The
Submission Of

2.1.2 Other Government documents, drawings, and publications. The following other Government documents, drawings, and publications form a part of this standard to the extent specified herein.

CODE OF FEDERAL REGULATIONS (CFR)

Title 29 - Department Of Labor, Occupational Safety And Health
Agency; General Industry Standards And Interpretations
Title 40 - Environmental Protection Agency; Hazardous Waste And
Consolidated Permit Regulations
Title 49 - Department Of Transportation; Hazardous Materials Regu-
lations

DOD 4145.19-R-1 - Storage And Materials Handling
DOD 4160.21-M - Defense Utilization And Disposal Manual
DOD 6050.5-LR - DOD Hazardous Materials Information System, Hazardous
Item Listing
TB MED 502 - Occupational And Environmental Health
(DLAM 1000.2) Respiratory Protection Program
TB MED 506 - Occupational And Environmental Health
Occupational Vision

MIL-STD-1443

TM 38-250 - Packaging, Materials Handling - Preparation Of Hazardous
Materials For Military Air Shipment

NATIONAL INSTITUTE FOR OCCUPATIONAL SAFETY AND HEALTH (NIOSH)
Registry of Toxic Effects of Chemical Substances
Recommendation for Environmental Exposure Limits

(Copies of specifications, standards, handbooks, drawings, and publications required by contractors in connection with specific acquisition functions should be obtained from the contracting activity or as directed by the contracting officer.)

2.2 Other publications. The following document(s) form a part of this standard to the extent specified herein. Unless otherwise specified, the issues of the documents which are DOD adopted shall be those listed in the issue of DODISS specified in the solicitation. The issues of documents which have not been adopted shall be those in effect on the date of the cited DODISS.

AMERICAN CONFERENCE OF GOVERNMENT INDUSTRIAL HYGIENISTS (ACGIH)

TLVs Threshold Limit Values for Chemical Substances in the Work Environment Adopted by American Conference of Government Industrial Hygienists (ACGIH) with Intended Changes.

(Application for copies should be addressed to American Conference of Governmental Industrial Hygienists, 6500 Glenway Avenue, Bldg D-7, Cincinnati, OH 45211.)

ASTM

ASTM D 874 - Standard Test Method for Sulfated Ash From Lubricating
Oils And Additives

ASTM D 1209 - Standard Test Method for Color Of Clear Liquids (Platinum-
Cobalt Scale)

(Application for copies should be addressed to ASTM, 1916 Race Street, Philadelphia, PA 19103.)

NATIONAL FIRE PROTECTION ASSOCIATION (NFPA)

National Fire Codes

(Application for copies should be addressed to National Fire Protection Association, Battery March Park, Quincy, MA 02269.)

(Nongovernment standards are generally available for reference from libraries. They are also distributed among nongovernment standards bodies and using Federal agencies.)

2.3 Order of precedence. In the event of a conflict between the text of this standard and the references cited herein, the text of this standard shall take precedence.

MIL-STD-1443

3. DEFINITIONS (Not applicable)

MIL-STD-1443

4. GENERAL REQUIREMENTS

4.1 Packaging data and labeling. All chemicals included in this standard shall be packaged in accordance with Federal Specification PPP-C-2020 and all applicable documents referenced therein. Shipping containers shall be labeled in accordance with current Department of Transportation (DOT) Hazardous Materials Regulations applicable to each chemical when shipping by military aircraft the requirements of TM 38-250 shall apply. In addition, each item shall be packaged and labeled as specified in the applicable contract or order. All labels shall also comply with Hazard Communication Standard, 29 CFR 1910.1200 (f).

4.2 Hazardous materials information. DOD 6050.5-LR, DOD Hazardous Materials Information System (HMIS) acquires, reviews, stores, and disseminates Material Safety Data Sheet (MSDS) information for all hazardous materials used by DOD. The contractual acquisition of a MSDS is accomplished through use of Federal Acquisition Regulation, paragraph 52.223-3, Hazardous Material Identification and Material Safety Data. The MSDS is prepared in accordance with the instructions in FED-STD-313; and shall comply with requirements of Hazard Communication Standard, 29 CFR 1910.1200 (g).

4.3 Safety.

4.3.1 Personal protective measures. The necessary respiratory, eye and skin protection to be used when handling chemicals shall be prescribed by the responsible installation industrial hygiene, medical and safety authorities.

4.3.1.1 Respiratory protection. Respirators, approved by the National Institute for Occupational Safety and Health (NIOSH) or the Mine Safety and Health Administration (MSHA) or by particular respiratory schedules of the Bureau of Mines (BM) for the compounds being used, may be employed for intermittent exposure or for supplementing other control measures (refer to TB MED 502 or DLAM 1000.2). Ventilation shall be adequate to remove hazardous concentrations.

4.3.1.2 Skin protection. Personnel using these compounds shall be provided with and required to use impervious gloves, sleeves, aprons, and boots whenever indicated. Protective creams and ointments commonly known as "barrier creams" may be of value in certain cases. However, barrier creams shall not be used to replace protective clothing. In case of contact with the skin, wash affected areas thoroughly with water. Eye lavages and emergency showers shall be located where there is a potential for direct contact with harmful chemicals.

4.3.1.3 Face and eye protection. Personnel using these compounds shall be provided with and required to wear chemical splash-proof safety goggles. In addition, face shields shall be provided and worn over the goggles if splashing could occur. In case of contact with the eyes, immediately irrigate with copious amounts of water for at least 20-30 minutes, and obtain medical attention. (Refer to TB MED 506.)

4.3.1.4 Training. Training shall be provided to ensure that employees know potential hazards of the chemicals with which they come in contact and the symptoms of over-exposure as well as how these chemicals affect the body and bodily functions. A person or persons shall be adequately trained to render first aid.

MIL-STD-1443

4.3.1.5 Exercises. Participation in exercises shall be stressed to demonstrate skills in the use of personal protective equipment and emergency response equipment.

4.3.2 Storage conditions. DOD 4145.19-R-1 describes general storage practices and requirements for hazardous materials in the DOD supply system. Specific requirements provided in the following paragraphs are supplementary in nature and shall be observed in consonance with the DOD storage regulations.

4.3.2.1 Flammable, combustible, pyrophoric and ignitable materials. A flammable material is generally any solid, liquid, vapor or gas that ignites easily and burns rapidly. Combustible materials are generally those that are difficult to ignite and burn slowly. The DOT, in Part 173, Subpart D, Section 173.115 of 49 CFR, defines a flammable liquid as one having a closed cup flash point below 100°F (37.8°C). A combustible liquid is defined, by DOT in the above reference, as one having a closed cup flash point at or above 100°F and below 200°F (93.3°C). A pyrophoric liquid is defined, by DOT in the above reference, as one that ignites spontaneously in dry or moist air at or below 130°F (54.5°C). Materials with flash points of 200°F or higher are not considered to be nonflammable or noncombustible, but are to be considered as burnable. The Environmental Protection Agency (EPA), in Part 261, Subpart C, Section 261.21 of 40 CFR*, designates the criteria for flammable and combustible materials and oxidizers that exhibit the characteristic of ignitability (I). Liquids with closed cup flash points of less than 140°F (60°C) are defined by EPA as ignitable. The autoignition point (temperature) of a substance is generally defined as the minimum temperature required to initiate or cause self-sustained combustion in the absence of a spark or flame. Materials that ignite easily under normal industrial conditions are considered to be dangerous fire hazards. Such materials shall be stored in a manner to prevent ignition and combustion. Easily ignitable substances, such as reducing agents, shall be kept away from strong oxidizing agents. All containers shall be tightly sealed. It is important to provide adequate ventilation in storage areas, and to locate the storage areas of these items away from fire hazards. Ample fire-control equipment shall be easily accessible. Storage buildings, rooms and cabinets shall comply with provisions of the National Fire Codes. The building shall be electrically grounded and signs posted to prevent the lighting of matches or smoking in the area. Flammable storage areas shall be equipped with smoke or fire detection equipment.

4.3.2.2 Water-sensitive fire and explosive hazardous materials. These are materials that react on contact with water or steam to ignite or evolve heat or explosive gases. Such materials exhibit the characteristic of reactivity (R) as designated by the EPA in Section 261.23 of the above reference.* These materials shall be stored in well-ventilated, cool, dry areas. All containers shall be tightly sealed. These materials are a fire hazard in contact with water or moisture; therefore, it is essential that no sprinkler be used. Otherwise, the building shall conform to that required for storage of flammable materials. The building shall be waterproof, located on high ground, and separated from other storage areas.

4.3.2.3 Incompatible materials. Materials that are chemically incompatible shall be segregated in the storage of both serviceable and unserviceable items.

*Refers only to materials that have become waste materials.

MIL-STD-1443

The degree of segregation will depend upon DOD 4145.19-R-1 and local supplementary requirements that insure safe storage conditions. Hazardous storage compatibility codes are provided in the HMIS referred to in 4.2.

4.3.3 Chemical hazardous exposure limits. Chemical hazardous exposure limits for airborne concentrations of substances are obtained from the current TLVs® Threshold Limit Values for Chemical Substances in the Work Environment, adopted by the American Conference of Government Industrial Hygienists (ACGIH); current Occupational Safety and Health Administration (OSHA) Permissible Exposure Limits (PEL), 29 CFR, Section 1910.1000; and NIOSH Recommendation for Environmental Exposure Limits. Such information is also shown in MSDSs and the HMIS referred to in 4.2. Carcinogenic substances are listed by OSHA in Category I for confirmed carcinogens, and in Category II for suspected carcinogens. Category I substances have standard exposure limits set at the lowest possible levels. Category II substances have standard exposure limits set to prevent acute or chronic effects.

4.3.4 Toxicity. Toxicity information for chemical compounds is available from various publications and from MSDSs, which are collected in DOD 6050.5-LR Hazardous Materials Information System.

4.3.4.1 EPA Toxic (T). Some chemical compounds have been designated by the EPA as toxic (T) in accordance with the criteria shown in Part 261, Subpart E, Section 261.11(a)(3) of 40 CFR.* Some commercial chemical products are listed as toxic under Subpart D, Section 261.33(f).

4.3.4.2 EPA Acute Hazardous Toxicity (H). Some chemical compounds have been designated by the EPA as acute hazardous (H) in toxicity in accordance with the criteria shown in Subpart B, Section 261.11(a)(2) of the above reference.* Some commercial chemical products are listed as acute hazardous in toxicity under Subpart D, Section 261.33(e).

4.3.4.3 Hazardous toxic constituents. A list of chemical compounds and substances, shown to have toxic effects on humans or other life forms, is contained in Appendix VIII to 40 CFR Part 261; and the Registry of Toxic Effects of Chemical Substances.

4.4 Pollution and disposal.

4.4.1 Pollution potential. All items described in this standard shall be assumed to have a pollution potential. However, to minimize this potential, the proper use, storage and disposal methods shall be strictly followed.

4.4.2 Disposal of excess or unserviceable material. To minimize disposal problems, it is recommended that no more than a one year's supply of each item listed in this standard be stocked. When stocks have been declared excess or unserviceable, they will be disposed of in accordance with the Defense Utilization and Disposal Manual, DOD 4160.21-M, and applicable DOD Policy Memoranda. Guidance can be obtained from your servicing Defense Reutilization and Marketing Office (DRMO) on procedures required for proper reporting and turn-in.

*Refers only to materials that have become waste materials.

MIL-STD-1443

4.4.3 Disposal and storage of hazardous wastes. Items are classified and managed as hazardous wastes as defined by the Resource Conservation and Recovery Act (RCRA) (Public Law 94-580). Items have been identified as meeting the characteristics (i.e., ignitable, corrosive, reactive or EP toxic) or are listed (i.e., toxic or acute hazardous) according to Identification and Listing of Hazardous Waste, Part 261; 40 CFR; or have been determined to be hazardous wastes by declaration of the Defense Reutilization and Marketing Service (DRMS) in accordance with procedures set forth in DOD 4160.21-M. Disposal of such items shall be managed in accordance with the Installation Environmental Office, the DRMO, or the Safety and Health Office to insure proper reporting of disposal and treatment actions to the US EPA and State; and shall be managed in accordance with Federal, State and local laws. The three main disposal methods are turn-in to the DRMO, on-post disposal by installation personnel, or disposal by commercial contract. Hazardous wastes that cannot be used, or disposed of as stated in 4.4.3.2, shall be stored under environmentally safe conditions until suitable methods of disposal are determined. Short-term storage (less than 90 days) requires proper containment (i.e., packaging and facilities) in accordance with Section 262.34, Part 262 of the above reference. Long-term storage (greater than 90 days) requires permitting by the EPA or by the state under Public Law 94-580 (RCRA), in compliance with the requirements of 40 CFR Parts 264 and 265. Physical custody will be accomplished by the activity with conforming storage or most nearly conforming storage. When physical custody is in question, the Post Commander will make the final decision. In all cases where the wastes are to be collected, stored, transported and disposed of at a state or local permitted disposal facility, the identity and description of the waste shall be maintained and recorded in accordance with Part 262 of the above reference. Transportation of the waste shall be in accordance with Part 263 of the above reference, Standards Applicable to Transporters of Hazardous Waste.

4.4.3.1 Cleanup of liquid spills. To control the migration of spilled or leaking liquids, dike around the item with an inert, dry absorbent (e.g., clay, sawdust or vermiculite) or follow installations spill plans (Spill Prevention Control and Countermeasure Plan and Installations Spill Contingency Plan). Control entry to the spill site and segregate salvageable materials away from the spill area. Initiate waste cleanup operations immediately in accordance with local procedures. The residue shall be safely handled and transported to an approved or permitted disposal or storage facility. Packaging, labeling, transportation and record-keeping requirements for this waste material are determined by the appropriate Federal and State agencies and local procedures. It is recommended that all activities involving disposal preparation and transportation to commercial facilities be properly coordinated with the appropriate Federal and State agencies responsible for health and environmental aspects of hazardous materials. It is imperative that the proper description of waste accompany the packaged item at all times. Final disposal of the waste item shall be accomplished by reutilization, transfer, donation or sales by DRMS in accordance with DOD 4160.21-M or by ultimate disposal as described in 4.4.3.2. Spill residue, including contaminants, to be turned in to the DRMO shall first be properly identified, containerized, and labeled. For large scale spills that grossly contaminate the environment, the Chemical Transportation Emergency Center (CHEMTREC), can be called for assistance. Applicable procedures

MIL-STD-1443

of the local spill control plan shall be followed. Necessary respiratory, eye, and skin protection measures are to be used while performing cleanup operations.

4.4.3.2 Ultimate disposal. Ultimate disposal shall be accomplished at a permitted or approved hazardous waste treatment or disposal facility designated by the Installation Environmental Office, DRMO, or Safety and Health Offices.

4.4.4 DISCLAIMER. RECOMMENDED DISPOSAL INSTRUCTIONS ARE FORMULATED FOR USE BY ELEMENTS OF THE DEPARTMENT OF DEFENSE. THE UNITED STATES OF AMERICA IN NO MANNER WHATSOEVER EITHER EXPLICITLY OR IMPLICITLY WARRANTS, STATES, OR INTENDS SAID INSTRUCTION, TO HAVE ANY APPLICATION, USE OR VIABILITY BY OR TO ANY PERSON OR PERSONS CONTRACTING OUTSIDE THE DEPARTMENT OF DEFENSE OR ANY PERSON OR PERSONS CONTRACTING WITH ANY INSTRUMENTALITY OF THE UNITED STATES OF AMERICA AND DISCLAIMS ALL LIABILITY FOR SUCH USE. ANY PERSON USING THESE INSTRUCTIONS WHO IS NOT A MILITARY OR CIVILIAN EMPLOYEE OF THE UNITED STATES OF AMERICA SHOULD SEEK COMPETENT PROFESSIONAL ADVICE TO VERIFY AND ASSUME RESPONSIBILITY FOR THE SUITABILITY OF THESE INSTRUCTIONS TO THEIR PARTICULAR SITUATION REGARDLESS OF SIMILARITY TO A CORRESPONDING DEPARTMENT OF DEFENSE OR OTHER GOVERNMENT SITUATION.

MIL-STD-1443

5. DETAIL REQUIREMENTS

- 5.1 Name. ARSENIC PENTOXIDE, ANALYZED REAGENT As₂O₅ FW 229.84
 Arsenic anhydride
 Arsenic (V) oxide

5.1.1 Technical Description. Arsenic pentoxide (pure) exists as white amorphous, deliquescent crystals with a density of 4.32 g/cm³. It has a melting point of 315°C with decomposition. Its solubility in water is 150 g/cm³ at 16°C and 76.7 g/cm³ at 100°C with the formation of arsenic acid. It is soluble in alcohol.

5.1.2 Specification. Military, MIL-A-51313, Arsenic Pentoxide, Analyzed Reagent.

5.1.3 Requirements. The Military Specification requirements for arsenic pentoxide, analyzed reagent is shown in Table I.

TABLE I. Arsenic pentoxide, analyzed reagent - requirements.

Assay (as As ₂ O ₅), min, % by wt	99.0
Maximum Limits of Impurities (% by wt)	
Arsenic trioxide (As ₂ O ₃)	0.05
Chloride (Cl)	0.005
Heavy metals (as Pb)	0.010
Iron (Fe)	0.010
Nitrate (NO ₃)	To pass color test

5.1.4 Use. Arsenic Pentoxide, analyzed reagent is intended for use as a laboratory reagent.

5.1.5 Safety. Arsenic pentoxide is highly toxic. Acute poisoning usually results from ingestion; Chronic poisoning from either ingestion or inhalation. Acute poisoning results in marked irritation of the stomach and intestines with nausea, vomiting and diarrhea. Chronic poisoning whether through ingestion or inhalation may result in disturbances of the digestive system. Liver damage may also occur. Inhalation of dust or mist must be avoided. Contact with the eyes, skin and clothing must be avoided. Wash thoroughly after handling. Arsenic pentoxide be used with adequate ventilation. The TLV for arsenic and soluble compounds as As is 0.2 mg/m³. The PEL is 10 mg/m³. For personal protective measures, refer to section 4.3.1.

If heated to decomposition arsenic pentoxide emits highly toxic fumes. It can form AsH₃ by reduction with zinc in acid solution or by reaction with hydrogen gas.

5.1.6 Storage. Arsenic pentoxide must be stored in cool, dry, well ventilated place in tightly closed containers away from fire hazards, acids and acid fumes and combustible materials. Refer to section 4.3.2.

5.1.7 Disposal. For appropriate procedures, contact the Installation Environmental Office, the DRMO, or Safety and Health Offices. Refer to section 4.4.

MIL-STD-1443

Arsenic pentoxide has an EPA Hazardous Waste Classification - EP Toxic, Waste Number D004; and Acute Hazardous Toxic, Waste Number P011.

5.2 Name. ARSENIC PENTOXIDE, n-HYDRATE, ANALYZED REAGENT

$\text{As}_2\text{O}_5 \cdot n\text{H}_2\text{O}$ FW 229.84+nH₂O

Arsenic (V) oxide, n-hydrate

5.2.1 Technical description. Arsenic pentoxide (pure) exists as white amorphous, deliquescent crystals with a density of 4.32 g/cm³. Its melting point of 315°C with decomposition. Its solubility in water is 150 g/100cm³ at 16°C and 76.7 g/100cm³ at 100°C with formation of arsenic acid. It is soluble in alcohol.

5.2.2 Specification. None

5.2.3 Requirements. Arsenic pentoxide, n-hydrate, analyzed reagent is commercially available with requirements as shown in Table II.

TABLE II. Arsenic pentoxide, n-hydrate analyzed reagent - requirements.

Assay, (as As_2O_5)(dried basis) (by iodometry) min, % by wt	97.0
Maximum Limits of Impurities (% by wt)	
Loss on heating at 350°C	15.0
Insoluble in NH_4OH	0.05
Chloride (Cl)	0.002
Nitrate (NO_3)	0.005
Sulfides (S)	0.001
Arsenic trioxide (As_2O_3)	0.05
Iron (Fe) (by AAS)	0.05
Lead (Pb) (by AAS)	0.05

5.2.4 Use. Arsenic pentoxide, n-hydrate, analyzed reagent is intended for use as a laboratory reagent.

5.2.5 Safety. Arsenic pentoxide, n-hydrate is highly toxic. Acute poisoning usually results from ingestion; chronic poisoning from either ingestion or inhalation. Acute poisoning results in marked irritation of the stomach and intestines with nausea, vomiting and diarrhea. Chronic poisoning by any route may result in disturbances of the digestive system. Liver damage may also occur. Inhalation of dust or mist must be avoided. Contact with the eyes, skin and clothing must be avoided. Wash thoroughly after handling. Arsenic pentoxide, n-hydrate must be used with adequate ventilation. Arsenic pentoxide, n-hydrate is a confirmed carcinogen. The TLV for arsenic and soluble compounds as As is 0.2 mg/m³. The PEL is 10 mg/m³. For personal protective measures, refer to section 4.3.1.

If heated to decomposition arsenic pentoxide emits highly toxic fumes. It can form AsH_3 by reduction with zinc in acid solution or by reaction with hydrogen gas.

5.2.6 Storage. Arsenic pentoxide, n-hydrate must be stored in a cool, dry, well ventilated place in tightly closed containers away from fire hazards, acids, acid fumes and combustible materials. Refer to section 4.3.2.

MIL-STD-1443

5.2.7 Disposal. For appropriate procedures, contact the Installation Environmental Office, the DRMO, or Safety and Health Offices. Refer to section 4.4.

Arsenic pentoxide has an EPA Hazardous Waste Classification - EP Toxic, Waste Number D004; Acute Hazardous Toxic, Waste Number P011.

5.3 Name. ARSENIC TRIOXIDE, ANALYZED REAGENT. As_2O_3 FW 197.84
 White arsenic
 Arsenous oxide
 Arsenic III trioxide

5.3.1 Technical description. Arsenic trioxide (pure) exists as amorphous or vitreous crystals with a density of 3.738 g/cm^3 . Its melting point is 315°C . Its solubility in water is 3.7 g/100cm^3 at 20°C and 10.14 g/100cm^3 at 100°C with the formation of arsenous acid. It sublimates at 193°C .

5.3.2 Specification. Federal, O-C-265, Chemicals, Analytical; General Specification For.

5.3.3 Requirements. The Federal Specification requirements for arsenic trioxide, analyzed reagent is shown in Table III.

TABLE III. Arsenic trioxide, analyzed reagent - requirements (ACS).

Maximum Limits of Impurities (% by wt)	
Residue after ignition	0.02
Insoluble in dilute hydrochloric acid	0.01
Chloride (Cl)	0.005
Sulfide (S), to pass color test, about	0.001
Antimony (Sb)	0.002
Lead (Pb)	0.001
Maximum Limits of Impurities (ppm)	
Iron (Fe)	5.0

Arsenic trioxide, analyzed reagent is commercially available with a minimum purity of 99.95%.

5.3.4 Use. Arsenic trioxide, analyzed reagent is intended for use as a reagent in analytical chemistry.

5.3.5 Safety. Arsenic trioxide is highly toxic. The acute symptoms of exposure, headache, dizziness, garlic odor breath, numbness, chills and tingling of hands and feet, are usually latent for about two days. The chronic symptoms usually appear one to six weeks after onset of exposure. The skin becomes discolored, dry and scaly. Other symptoms may include edema of eyelids, corneal necrosis, nasal irritation and dryness of throat. The central nervous system and gastrointestinal tract may also be affected. Inhalation of dust and mist of arsenic trioxide must be avoided. Wash thoroughly after handling. Arsenic trioxide must be used with adequate ventilation. Arsenic trioxide is a confirmed carcinogen. The TLV for arsenic and soluble compounds as As is 0.2 mg/m^3 . The PEL is 10 mg/m^3 . For personal protective measures, refer to section 4.3.1.

MIL-STD-1443

5.3.6 Storage. Arsenic trioxide must be stored in a cool, dry, well ventilated place in tightly closed containers away from food, food products and combustible materials. Protect container against physical damage. Refer to section 4.3.2.

5.3.7 Disposal. For appropriate procedures, contact the Installation Environmental Office, the DRMO, or Safety and Health Offices. Refer to section 4.4.

Arsenic trioxide has an EPA Hazardous Waste Classification - EP Toxic, Waste Number D004; and Acute Hazardous Toxic, Waste Number P012.

5.4 Name. BORIC ACID, ANALYZED REAGENT. H_3BO_3 FW 61.83
Boracic acid
Orthoboric acid

5.4.1 Technical description. Boric acid (pure) exists as colorless triclinic crystals with a density of 1.435 g/cm^3 . Its solubility in water is 6.35 g/100cm^3 at 20°C and 27.6 g/100cm^3 at 100°C . It is also soluble in glycerol (20 g/100cm^3 at 25°C). Aqueous solutions of H_3BO_3 have approximately pHs of 5.4 at 0.5%, 5.1 at 1.0%, 4.6 at 2.0%, 4.2 at 3.0%, 3.9 at 4.0%, and 3.7 at 4.65% by weight concentrations. On heating to 169° to 175°C , H_3BO_3 loses water to form metaboric acid HBO_2 . Further heating forms $H_2B_4O_7$ and the oxide B_2O_3 .

5.4.2 Specification. Federal, O-C-265, Chemicals, Analytical; General Specification For.

5.4.3 Requirements. The Federal Specification requirements for boric acid, analyzed reagent is shown in Table IV.

TABLE IV. Boric acid, analyzed reagent - requirements (ACS).

Maximum Limits of Impurities (% by wt)	
Insoluble in methanol	0.005
Nonvolatile with methanol	0.05
Chloride (Cl)	0.001
Phosphate (PO_4)	0.001
Sulfate (SO_4)	0.010
Calcium (Ca)	0.005
Heavy metals (as Pb)	0.001
Iron (Fe)	0.001
Maximum Limits of Impurities (ppm)	
Arsenic (As)	1.0

Boric acid, analyzed reagent is commercially available in granular and powdered forms with a minimum purity, as H_3BO_3 , >99.5%.

5.4.4 Use. Boric acid, analyzed reagent is used in medicine.

5.4.5 Safety. Boric acid is not considered to be an industrial poison. It is not absorbed through intact skin, but is absorbed through mucous membranes, burned or wounded skin area, or if ingested. The fatal dose if ingested is about 15-20g for an adult. Absorption will produce pathological

MIL-STD-1443

changes. Inhalation of dust must be avoided. Contact with abraded skin must be avoided. For personal protective measures, refer to section 4.3.1.

5.4.6 Storage. Boric acid must be stored in a cool, dry place in tightly closed containers. Refer to section 4.3.2.

5.4.7 Disposal. For appropriate procedures, contact the Installation Environmental Office, the DRMO, or Safety and Health Offices. Refer to section 4.4.

5.5 Name. CHLOROSULFONIC ACID, REAGENT. ClSO₃H FW 116.52
Chlorosulfuric acid

5.5.1 Technical description. Chlorosulfonic acid (pure) exists as a colorless fuming liquid with a specific gravity of 1.766 at 18°C. It has a melting point of -80°C and a boiling point of 158°C. It decomposes to H₂SO₄ and HCl in cold water and decomposes in alcohol. It is insoluble in CS₂.

5.5.2 Specification. Military, MIL-C-51130, Chemicals, Reagent Grade, General Specification For.

5.5.3 Requirements. The Military Specification requirements for chlorosulfonic acid, reagent is shown in Table V.

TABLE V. Chlorosulfonic acid, reagent - requirements.

Boiling point, °C	150-153
Specific gravity, 20°/4°C	1.760

Chlorosulfonic acid, reagent is commercially available with a minimum purity of 99.0%.

5.5.4 Use. Chlorosulfonic acid, reagent is intended for use as a laboratory reagent.

5.5.5 Safety. Chlorosulfonic acid is toxic, corrosive and nonflammable. It can cause severe acid burns and is very irritating to the eyes, lungs and mucous membranes. It can cause acute toxic effects either in the liquid or vapor state. Inhalation of concentrated vapor may cause serious damage to lung tissue. Contact with eyes can cause severe burns. It also causes skin irritation due to its highly corrosive action. Inhalation of vapor or mist must be avoided. Wash thoroughly after handling. Chlorosulfonic acid must be used with adequate ventilation. For personal protective measures, refer to section 4.3.1.

Chlorosulfonic acid reacts violently with water. When exposed to the atmosphere, chlorosulfonic acid fumes release hydrochloric acid fumes and sulfuric acid mist by reacting with moisture in the air. Inhalation of these fumes or mist may also cause lung injury. The TLV (ceiling limit) for hydrogen chloride is 5 ppm (7mg/m³). The TLV for sulfuric acid is 1 mg/m³. The PEL is the same. Chlorosulfonic acid creates hazardous exothermic reaction with many organic and combustible materials, nitrates, chlorates, carbides, sulfides, cyanides and metallic powders.

MIL-STD-1443

5.5.6 Storage. Chlorosulfonic acid must be stored in cool, dry, well ventilated place in tightly closed containers away from incompatible materials such as organic and combustible materials, nitrates, chlorates, carbides, sulfides, cyanides and metallic powders. Do not allow water to enter containers and do not wash out containers or use for other purposes. Refer to section 4.3.2.

5.5.7 Disposal. For appropriate procedures, contact the Installation Environmental Office, the DRMO, or Safety and Health Offices. Refer to section 4.4.

Chlorosulfonic acid has an EPA Hazardous Waste Classification - Corrosive, Waste Number D002; Reactive, Waste Number D003.

5.6 Name. CHROMIUM TRIOXIDE, ANALYTICAL REAGENT. CrO_3 FW 99.99
 Chromic trioxide
 Chromic anhydride
 Chromic acid
 Chromium (VI) oxide

5.6.1 Technical description. Chromium trioxide (pure) exists as red rhombic deliquescent crystals with a density of 2.70 g/cm^3 and a melting point of 196°C . It decomposes to Cr_2O_3 and oxygen at 250°C . Chromium trioxide reacts with water in different proportions, forming chromic acid, H_2CrO_4 , dichromic acid, $\text{H}_2\text{Cr}_2\text{O}_7$, and trichromic acid, $\text{H}_2\text{Cr}_3\text{O}_{10}$.

5.6.2 Specification. Federal, O-C-265, Chemicals, Analytical, General Specification For.

5.6.3 Requirements. The Federal Specification requirements for chromium trioxide, analyzed reagent is shown in Table VI.

TABLE VI. Chromium trioxide, analyzed reagent - requirements (ACS).

Assay (CrO_3), min, % by wt	98.000
Maximum Limits of Impurities (% by wt)	
Insoluble matter	0.010
Chloride (Cl)	0.005
Nitrate (NO_3)	0.050
Sulfate (SO_4)	0.005
Sodium (Na)	0.200
Iron, aluminum, barium	0.030

5.6.4 Use. Chromium trioxide, analyzed reagent is intended for use as a reagent in analytical chemistry.

5.6.5 Safety. Chromium trioxide is highly toxic and a strong oxidizer. It is an irritant to eyes, skin and mucous membranes. The lesions are confined to the exposed parts, affecting chiefly the skin of the hands and forearms and the mucous membranes of the septum. Inhalation of dust or mist must be avoided. Contact with the eyes, skin and clothing must be avoided. Wash thoroughly after handling. Chromium trioxide must be used with adequate ventilation. Chromium trioxide is an OSHA Category I car-

MIL-STD-1443

cinogen. The TLV for chromium (VI) compounds as Cr is 0.05 mg/m³. For personal protective measures, refer to section 4.3.1.

Chromium trioxide decomposes when heated to 250°C, liberating oxygen to support combustion. It can cause violent reactions, ignition and explosion by contact with various organic materials or reducing materials especially, contact with acetic acid, analine, alcohol, acetone and grease. Containers may explode when involved in fire.

5.6.6 Storage. Chromium trioxide must be stored in cool, dry well ventilated place in tightly closed containers away from heat, sparks, open flame, combustible and easily oxidizable materials. Protect containers against physical damage. Refer to section 4.3.2.

5.6.7 Disposal. For appropriate procedures, contact the Installation Environmental Office, the DRMO, or Safety and Health Offices. In case of dry spills gather up and place in nonreactive containers. In case of small solution spills control with nonreactive absorbent. In case of large spills dike to prevent spreading. A frequently used method of disposal is to reduce Cr (VI) to Cr (III) and then to precipitate the reduced chromium as sludge with alkali. Where governing laws and regulations permit, the precipitated chromium sludge should be encased in a concrete block or other impervious material prior to disposal in a permitted landfill. Refer to section 4.4.

Chromium trioxide has an EPA Hazardous Waste Classification - EP Toxic, Waste Number D007; Ignitable, Waste Number D001.

5.7 Name. HYDROBROMIC ACID (48%), ANALYZED REAGENT. HBr FW 80.91
Hydrogen bromide

5.7.1 Technical description. Hydrobromic acid is a clear, colorless gas or a pale yellow liquid. It has a molecular weight of 80.92 g/mol. The specific gravity of the liquid is 2.7 (H₂O=1) at 20/4°C. It has a melting point of -88.5°C and a boiling point of -67.0°C. Its water solubility is 221 g/100cm³ at 0°C and 130 g/100cm³ at 100°C. A constant boiling solution is formed with a specific gravity of 1.49 (H₂O=1) at 20/4°C containing 48% hydrobromic acid. It has a melting point of -11°C and a boiling point of 126°C.

5.7.2 Specification. Federal, O-C-265, Chemicals, Analytical; General Specification For.

5.7.3 Requirements. The Federal Specification requirements for hydrobromic acid (48%), analyzed reagent is shown in Table VII.

TABLE VII. Hydrobromic acid (48%), analyzed reagent - requirements (ACS).

Assay, min-max, % by wt	47.0-49.0
Maximum Limits of Impurities (% by wt)	
Organic substance	To pass odor test
Residue after ignition	0.002
Chloride (Cl)	0.05

MIL-STD-1443

TABLE VII. Hydrobromic acid (48%), analyzed reagent - requirements (ACS)
- Continued.

Iodide (I) to pass color test, about	0.003
Phosphate (PO ₄)	0.001
Sulfate and sulfite (as SO ₄)	0.003
Maximum Limits of Impurities (ppm)	
Arsenic (As)	0.5
Heavy metal (as Pb)	5
Iron (Fe)	1
Selenium (Se), to pass color test, about	0.01

5.7.4 Use. Hydrobromic acid (48%) analyzed reagent is intended for use as a reagent in analytical chemistry.

5.7.5 Safety. Hydrobromic acid solution is highly acidic and nonflammable. It is a strong irritant to the eyes, skin and mucous membranes and respiratory organs. Inhalation of vapors and mist must be avoided. Contact with the eyes, skin and clothing must be avoided. Hydrobromic acid must be used with adequate ventilation. The TLV for hydrobromic acid is 3 ppm (10 mg/m³). The PEL is the same. For personal protective measures, refer to section 4.3.1.

Hydrobromic acid itself is not flammable, but will generate hydrogen gas when in contact with metals causing it to become flammable.

5.7.6 Storage. Hydrobromic acid (48%) must be stored in cool, dry well ventilated place in tightly closed containers away from oxidizing materials, metals, wood and textiles. Refer to section 4.3.2.

5.7.7 Disposal. For appropriate procedures, contact the Installation Environmental Office, the DRMO, or Safety and Health Offices. Refer to section 4.4.

Hydrobromic acid (48%) has an EPA Hazardous Waste Classification - Corrosive.

5.8 Name. HYDROCHLORIC ACID, ANALYZED REAGENT HCl FW 36.46
Chlorohydric acid
Hydrogen chloride

5.8.1 Technical description. Hydrochloric acid (pure) exists as a colorless liquid with a melting point of -114.8°C and a boiling point of -84.9°C. Its water solubility is 82.3 g/100cm³ at 0°C and 56.1 g/100cm³ at 60°C. It is soluble in ether and benzene.

5.8.2 Specification. Federal, O-C-265, Chemical, Analytical; General Specification For.

5.8.3 Requirements. The Federal Specification requirements for hydrochloric acid, analyzed reagent is shown in Table VIII

MIL-STD-1443

TABLE VIII. Hydrochloric acid, analyzed reagent - requirements (ACS)

Appearance	Free from suspended matter or sediment
Assay (HCl), min-max, % by wt	36.5-38.0
Color (APHA) ^{1/} , max	10.0
Maximum Limits of Impurities (% by wt)	
Bromide (Br)	0.005
Maximum Limits of Impurities (ppm)	
Residue after ignition	5
Sulfate (SO ₄)	1
Sulfite (SO ₃)	1
Extractable organic substances, to pass impurities test, about	5
Free chlorine (Cl), to pass color test	about 1
Ammonium (NH ₄)	3
Arsenic (As)	0.01
Heavy metals (as Pb)	1
Iron (Fe)	0.2

^{1/} APHA requirements are the same as platinum-cobalt scale used in ASTM, designation D 1209.

5.8.4 Use. Hydrochloric acid, analyzed reagent is intended for use as a reagent in analytical chemistry.

5.8.5 Safety. Hydrochloric acid is a corrosive liquid. It is irritating to the eye, skin, mucous membranes and respiratory tract. Inhalation of vapor and mist must be avoided. Contact with the eyes, skin and clothing must be avoided. Hydrochloric acid must be used with adequate ventilation. The TLV (ceiling limit) for hydrochloric acid is 5 ppm (7 mg/m³). The PEL is the same. For personal protective measures, refer to section 4.3.1.

5.8.6 Storage. Hydrochloric acid must be stored in a cool, well ventilated place in tightly closed containers away from oxidizing materials. Protect containers against physical damage. Refer to section 4.3.2.

5.8.7 Disposal. For appropriate procedures, contact the Installation Environmental Office, the DRMO, or Safety and Health Offices. Refer to section 4.4.

Hydrochloric acid has an EPA Hazardous Waste Classification - Corrosive, Waste Number D002.

5.9 Name. HYDROFLUORIC ACID, ANALYZED REAGENT. HF FW 20.01
Hydrogen fluoride
Fluorohydric acid

5.9.1 Technical description. Hydrofluoric acid (pure) exists as a colorless, fuming liquid with a specific gravity of 0.987 at 13.6°C (H₂O=1). Its melting point is -83.1°C and boiling point is 19.5°C. It is infinitely soluble in cold water and very soluble in hot water.

MIL-STD-1443

5.9.2 Specification. Federal, O-C-265, Chemicals, Analytical; General Specification For.

5.9.3 Requirements. The Federal Specification requirements for hydrofluoric acid, analyzed reagent is shown in Table IX.

TABLE IX. Hydrofluoric acid, analyzed reagent - requirements (ACS).

Assay (HF), min-max (% by wt)	48.0-51.0
Maximum Limits of Impurities (% by wt)	
Fluosilicic acid (H ₂ SiF ₆)	0.01
Maximum Limits of Impurities (% by wt)	
Residue after ignition	5
Chloride (Cl)	5
Phosphate (PO ₄)	1
Sulfate and sulfite (as SO ₄)	5
Arsenic (As)	0.05
Copper (Cu)	0.1
Heavy metals (as Pb)	0.5
Iron (Fe)	1

5.9.4 Use. Hydrofluoric acid, analyzed reagent is intended for use as a reagent in analytical chemistry.

5.9.5 Safety. Hydrofluoric acid is a corrosive, noncombustible liquid. It is extremely irritating to the eyes, skin and mucous membranes. Inhalation of vapor or mist must be avoided. Inhalation of vapor or mist causes ulcers of the upper respiratory tract, and concentrations of 50 to 250 ppm are dangerous even for brief exposures. Contact with the eyes, skin, and clothing must be avoided. Hydrofluoric acid produces severe skin burns and may affect the subcutaneous tissues. Hydrofluoric acid must be used with adequate ventilation. The TLV (ceiling limit) for hydrogen fluoride (as F) is 3 ppm (2.5 mg/m³). For personal protective measures, refer to section 4.3.1.

Hydrofluoric acid is dangerous when heated, emitting corrosive fluoride fumes. It will react with steam to produce toxic and corrosive fumes. It is difficult to contain as it corrodes most substances except lead, wax, polyethylene and platinum.

5.9.6 Storage. Hydrofluoric acid must be stored in cool, well ventilated place in tightly closed containers made of resistant material such as polyethylene away from heat and sources of direct sunlight. Refer to section 4.3.2.

5.9.7 Disposal. For appropriate procedures, contact the Installation Environmental Office, the DRMO, or Safety and Health Offices. Refer to section 4.4.

Hydrofluoric acid has an EPA Hazardous Waste Classification - Corrosive; Toxic, Waste Number U134.

5.10 Name. IODIC ACID, ANALYZED REAGENT.

HI O₃ FW 175.91

MIL-STD-1443

5.10.1 Technical description. Iodic acid (pure) exists as colorless or pale yellow crystalline powder or rhombic crystals with a density of 4.629 g/cm³. It decomposes at 110°C. Its solubility in water is 286 g/100cm³ at 0°C and 473 g/100cm³ at 80°C. It is very soluble in 87% alcohol, slightly soluble in HNO₃ and insoluble in absolute alcohol, ether and chloroform. It is sensitive to light.

5.10.2 Specification. Federal, O-C-265, Chemicals, Analytical; General Specification For.

5.10.3 Requirements. The Federal Specification requirements for iodic acid, analyzed reagent is shown in Table X.

TABLE X. Iodic acid, analyzed reagent - requirements (ACS).

Assay (HIO ₃), min (% by wt)	99.50
Maximum Limits of Impurities (% by wt)	
Insoluble matter	0.01
Residue after ignition	0.02
Chloride and bromide (as Cl)	0.02
Iodide (I)	0.01
Nitrogen compounds (as N)	0.1
Sulfate (SO ₄)	0.015
Heavy metals (as Pb)	0.001
Maximum Limits of Impurities (ppm)	
Iron (Fe)	5

5.10.4 Use. Iodic acid, analyzed reagent is intended for use as a reagent in analytical chemistry. It is also used in medicine.

5.10.5 Safety. Iodic acid is a strong oxidizer and irritating to the eyes, skin and mucous membranes. Inhalation of dust or mist must be avoided. Contact with eyes, skin and clothing must be avoided. Iodic acid must be used with adequate ventilation. For personal protective measures, refer to section 4.3.1.

Iodic acid is a dangerous fire hazard in contact with combustible materials. If heated to decomposition, iodic acid can emit highly toxic fumes.

5.10.6 Storage. Iodic acid must be stored in cool, dry, well ventilated place in tightly closed containers away from combustible materials. Protect from light. Refer to section 4.3.2.

5.10.7 Disposal. For appropriate procedures, contact the Installation Environmental Office, the DRMO, or Safety and Health Offices. Refer to section 4.4.

5.11 Name. IODIC PENTOXIDE, ANALYZED REAGENT. I₂O₅ FW 333.81
Iodic anhydride

5.11.1 Technical description. Iodic pentoxide (pure) exists as white trimetric hygroscopic crystals with a density of 4.799 g/cm³ at 25/4°C. It decomposes between 300-350°C. It has a solubility of 187.4 g/cm³ in water

MIL-STD-1443

at 13°C and is very soluble in hot water. It is insoluble in alcohol, ether, chloroform and carbon disulfide. It is sensitive to light.

5.11.2 Specification. Federal, O-C-265, Chemicals, Analytical; General Specification For.

5.11.3 Requirements. The Federal Specification requirements for iodic pentoxide, analyzed reagent is shown in Table XI.

TABLE XI. Iodic pentoxide, analyzed reagent - requirements (USP).

Assay (I_2O_5), min, % by wt	98.5
Maximum Limits of Impurities (% by wt)	
Residue on ignition	0.05
Heavy metal (as Pb)	0.001
Iron (Fe)	0.002

5.11.4 Use. Iodic pentoxide, analyzed reagent is intended for use as a analytical chemistry.

5.11.5 Safety. Iodic pentoxide is an oxidizer and highly toxic. It is irritating to the eyes, skin and mucous membranes. Inhalation of vapor and dust must be avoided. Contact with the eyes, skin and clothing must be avoided. Iodic pentoxide must be used with adequate ventilation. Its vapor effect upon the body is similar to that of chlorine and bromine, but it is more irritating to the lungs. Signs and symptoms are irritation and burning of the eyes and lachrymation, cough, irritation of the nose and throat. For personal protective measures, refer to section 4.3.1.

Iodic pentoxide is a slight explosive hazard when exposed to heat or flame. There is a dangerous fire hazard from spontaneous chemical reaction. Iodine compound will react with steam to produce toxic fumes and can react vigorously with reducing materials.

5.11.6 Storage. Iodic pentoxide must be stored in a cool, dry, well ventilated place in tightly closed containers away from sources of heat and open flame and reducing materials. Protect from light. Refer to section 4.3.2.

5.11.7 Disposal. For appropriate procedures, contact the Installation Environmental Office, the DRMO, or Safety and Health Offices. Refer to section 4.4.

Iodic pentoxide has an EPA Hazardous Waste Classification - Ignitable, Waste Number D001.

5.12 Name. MOLYBDENUM TRIOXIDE, ANALYZED REAGENT. MoO_3 FW 143.94
Molybdic acid anhydride
Molybdenum (VI) oxide

5.12.1 Technical description. Molybdenum trioxide (pure) exists as colorless or white-yellow rhombic crystals with a density of 4.692 g/cm³ at 21°C. Its melting point is 795°C and it sublimates at 1155°C at 760 mm Hg.

MIL-STD-1443

Its solubility in water is 0.1066 g/100cm³ at 18°C and 2.055 g/cm³ at 70°C. It will precipitate in the presence of acids, and is soluble in bases.

5.12.2 Specification. Federal, O-C-265, Chemicals, Analytical; General Specification For.

5.12.3 Requirements. The Federal Specification requirements for molybdenum trioxide, analyzed reagent is shown in Table XII.

TABLE XII. Molybdenum trioxide, analyzed reagent - requirements (ACS).

Assay (MoO ₃), min, % by wt	99.5
Maximum Limits of Impurities (% by wt)	
Insoluble in dilute ammonium hydroxide	0.01
Chloride (Cl)	0.002
Nitrate (NO ₃) to pass color test	limit about 0.003
Arsenate, phosphate, and silicate (as SiO ₂)	0.001
Sulfate (PO ₄)	0.02
Ammonium (NH ₄)	0.002
Heavy Metals (as Ag)	0.005
Maximum Limits of Impurities (ppm)	
Phosphate	5

5.12.4 Use. Molybdenum trioxide, analyzed reagent is intended for use as a reagent in analytical chemistry.

5.12.5 Safety. Molybdenim trioxide is relatively nontoxic. It may be irritating to the eyes, skin and mucous membranes. Inhalation of dust or mist must be avoided. Contact with the eyes, skin and clothing must be avoided. Molybdenum trioxide must be used with adequate ventilation. The TLV for molybdenum trioxide is 10 mg/m³. The PEL is 15 mg/m³. For personal protective measures, refer to section 4.3.1.

5.12.6 Storage. Molybdenum trioxide must be stored in cool, dry, well ventilated place in tightly closed containers. Refer to section 4.3.2.

5.12.7 Disposal. For appropriate procedures, contact the Installation Environmental Office, the DRMO, or Safety and Health Offices. Refer to section 4.4.

5.13 Name. NITRIC ACID, ANALYZED REAGENT. HNO₃ FW 63.01
 Aqua fortis
 Hydrogen nitrate
 Azotic acid

5.13.1 Technical description. Nitric acid (pure) exists as a colorless liquid with a density of 1.5027 g/cm³ at 25/4°C. It freezes to a white solid at -42°C and boils at 83°C. It is soluble in all proportions in both hot and cold water. It decomposes in alcohol and is soluble in ether. An aqueous solution containing 68% of the acid has a constant boiling point of 120.5°C.

MIL-STD-1443

5.13.2 Specification. Federal, O-C-265, Chemicals, Analytical; General Specification For.

NOTE: Nitric acid, meeting ACS Specifications for reagent chemical is available as 69% and 90% (sometimes referred to as fuming).

5.13.3 Requirements. The Federal Specification requirements for nitric acid (69%) analyzed reagent is shown in Table XIII.

TABLE XIII. Nitric acid (69%) analyzed reagent - requirements (ACS).

Appearance	Colorless and free from suspended matter or sediment
Assay (HNO ₃), min-max, % by wt	69.0-71.0
Maximum Limits of Impurities (ppm)	
Residue after ignition	5
Chloride (Cl)	0.5
Sulfate (SO ₄)	1
Arsenic (As)	0.01
Heavy metal (as Pb)	0.2
Iron (Fe)	0.2

The Federal Specification requirements for nitric acid (90%) analyzed reagent is shown in Table XIV.

TABLE XIV. Nitric acid (90%) analyzed reagent - requirements (ACS).

Assay (HNO ₃), min, % by wt	90.0
Maximum Limits of Impurities (% by wt)	
Dilution test	to pass turbidity or precipitate test
Residue after ignition	0.002
Dissolved oxides (as N ₂ O ₃), to pass test, limit about	0.1
Maximum Limits of Impurities (ppm)	
Chloride (Cl)	0.7
Sulfate (SO ₄)	5
Arsenic (As)	0.3
Heavy metal (as Pb)	5
Iron (Fe)	2

5.13.4 Use. Nitric acid, analyzed reagent is intended for use as a reagent in analytical chemistry including the determination of trace metals

5.13.5 Safety. Nitric acid is a strong oxidizer and corrosive. It is very irritating to the eyes, skin and mucous membranes. Inhalation of vapors or mist must be avoided. Contact with the eyes, skin and clothing must be avoided. Nitric acid must be used with adequate ventilation. The TLV for nitric acid is 2 ppm (5 mg/m³). The PEL is the same. For personal protective measures, refer to section 4.3.1.

MIL-STD-1443

Nitric acid exhibits the properties of a strong acid and a strong oxidizing agent. Nitric acid is noncombustible but dangerously reactive with many materials. It can react violently with metallic powders, carbides, charcoal and hydrogen sulfide. It increases the flammability of combustible and readily oxidizable materials. It is a moderate fire hazard with reducing materials. When heated to decomposition it can emit toxic and corrosive oxides of nitrogen fumes.

5.13.6 Storage. Nitric acid must be stored in a cool, dry, well ventilated place in tightly closed acid resistant, nonreactive containers away from sources of heat, direct sunlight and oxidizable material. Protect container against physical damage. Refer to section 4.3.2.

5.13.7 Disposal. For appropriate procedures, contact the Installation Environmental Office, the DRMO, or Safety and Health Offices. Refer to section 4.4.

Nitric acid (over 40%) has EPA Hazardous Waste Classification - Ignitable, Waste Number D001; Corrosive, Waste Number D002.

5.14 Name. PHOSPHOMOLYBDIC ACID, ANALYZED, REAGENT.

$12\text{MoO}_3 \cdot \text{H}_3\text{PO}_4 \cdot x\text{H}_2\text{O}$ FW 1825.24+H₂O

Molybdophosphoric acid

Phospho-12-molybdic acid

5.14.1 Technical description. Phosphomolybdic acid exists as yellow tetragonal crystals with a melting point of 78-90°C. It is soluble in hot and cold water. It is very soluble in alcohol and ether.

5.14.2 Specification. Federal, O-C-265, Chemicals, Analytical; General Specification For.

5.14.3 Requirements. The Federal Specification requirements for phosphomolybdic acid, analyzed reagent is shown in Table XV.

TABLE XV. Phosphomolybdic acid, analyzed reagent - requirements (ACS).

Maximum Limits of Impurities (% by wt)	
Insoluble matter	0.01
Chloride (Cl)	0.02
Sulfate (SO ₄)	0.025
Ammonium (NH ₄)	0.01
Calcium (Ca)	0.02
Heavy metals (as Pb)	0.005
Iron (Fe)	0.005

5.14.4 Use. Phosphomolybdic acid, analyzed reagent is intended for use as a reagent in analytical chemistry for the determination of alkaloids and proteins. It is also used as a reagent in microscopy.

5.14.5 Safety. Phosphomolybdic acid is an oxidizer and corrosive solid. It is irritating to the eyes, skin and mucous membranes. Inhalation of dust or mist of phosphomolybdic acid must be avoided. Contact with the eyes, skin and clothing must be avoided. Phosphomolybdic acid must be used

MIL-STD-1443

with adequate ventilation. The TLV for molybdenum soluble compounds (as Mo) is 5 mg/m³. The PEL is the same. For personal protective measures, refer to section 4.3.1.

5.14.6 Storage. Phosphomolybdic acid must be stored in a cool, dry, well ventilated place in tightly closed containers away from sources of heat and open flame and oxidizable material. Refer to section 4.3.2.

5.14.7 Disposal. For appropriate procedures, contact the Installation Environmental Office, the DRMO, or Safety and Health Offices. Refer to section 4.4.

Phosphomolybdic acid has an EPA Hazardous Waste Classification - Ignitable, Waste Number D001; Corrosive, Waste Number D002.

5.15 Name. PHOSPHORIC ACID (META), ANALYZED REAGENT.

HPO₃ FW 79.98

Glacial phosphoric acid

5.15.1 Technical description. Phosphoric acid (meta) (pure) exists as colorless, deliquescent crystals with a density of 2.2-2.5 g/cm³. It sublimes on melting. It is very slowly soluble in cold water changing to H₃PO₄, the change is hastened by heating. It is soluble in alcohol and insoluble in liquid CO₂.

5.15.2 Specification. Federal, O-C-265, Chemicals Analytical, General Specification For.

NOTE: This reagent contains as a stabilizer a somewhat greater proportion of sodium metaphosphate than that corresponding to the empirical formula NaH(PO₃)₂.

5.15.3 Requirements. The Federal Specification requirements for phosphoric acid (meta), analyzed reagent is shown in Table XVI.

TABLE XVI. Phosphoric acid (meta), analyzed reagent - requirements (ACS).

Assay (HPO ₃), min-max, % by wt	34.0-36.0
Stabilizer, min-max, % by wt	58.0-62.0
Maximum Limits of Impurities (% by wt)	
Chloride (Cl)	0.001
Nitrate (NO ₃)	0.001
Sulfate (SO ₄)	0.005
Heavy metals (as Pb)	0.005
Iron (Fe)	0.005
Substances reducing permanganate (as H ₃ PO ₃)	0.02
Maximum Limits of Impurities (ppm)	
Arsenic (As)	1

5.15.4 Use. Phosphoric acid (meta), analyzed reagent is intended for use as a reagent in analytical chemistry.

5.15.5 Safety. Phosphoric acid (meta), is corrosive. It is irritating to the eyes, skin and mucous membranes. Inhalation of dust or mist must be avoided. Contact with the eyes, skin and clothing must be avoided. Phos-

MIL-STD-1443

phoric acid must be used with adequate ventilation. The TLV for phosphoric acid is 1 mg/m³. The PEL is the same. For personal protective measures, refer to section 4.3.1.

If heated to decomposition, phosphoric acid (meta) emits toxic oxides of phosphorus fumes.

5.15.6 Storage. Phosphoric acid (meta) must be stored in cool, dry, well ventilated place in tightly closed containers. Refer to section 4.3.2.

5.15.7 Disposal. For appropriate procedures, contact the Installation Environmental Office, the DRMO, or Safety and Health Offices. Refer to section 4.4.

Phosphoric acid has an EPA Hazardous Waste Classification - Corrosive, Waste Number D002.

5.16 Name. PHOSPHORIC ACID (ORTHO) ANALYZED REAGENT H₃PO₄ FW 98.00
Phosphoric acid

5.16.1 Technical description. Phosphoric acid (ortho) (pure) exists as a colorless liquid or as rhombic, deliquescent crystals with a density of 1.834 g/cm³ at 18°C. Its melting point is 42.35°C. Its solubility in cold water is 548 g/100cm³ and it is very soluble in hot water. It is soluble in alcohol.

5.16.2 Specification. Federal, O-C-265, Chemicals, Analytical, General Specification For.

5.16.3 Requirements. The Federal Specification requirements for phosphoric acid (ortho), analyzed reagent is shown in Table XVII.

TABLE XVII. Phosphoric acid (ortho), analyzed reagent - requirements (ACS).

Assay (H ₃ PO ₄), min, % by wt	85.0
Color (APHA), max	10.0
Maximum Limits of Impurities (% by wt)	
Insoluble matter, calcium, magnesium and ammonium hydroxide precipitate	0.005
Sulfate (SO ₄)	0.003
Volatile acids (as CH ₃ COOH)	0.001
Heavy metals (as Pb)	0.001
Iron (Fe)	0.003
Potassium (K)	0.005
Sodium (Na)	0.025
Reducing substances	To pass color test
Maximum Limits of Impurities (ppm)	
Chloride (Cl)	3.000
Manganese (Mn)	5
Nitrate (NO ₃)	5.000
Arsenic (As)	1.000

MIL-STD-1443

5.16.4 Use. Phosphoric acid (ortho), analyzed reagent is intended for use as a reagent in analytical chemistry.

5.16.5 Safety. Phosphoric acid (ortho) is moderately toxic and irritating to eyes, skin and mucous membranes. Inhalation of dust or mist must be avoided. Contact with the eyes, skin and clothing must be avoided. Phosphoric acid (ortho) must be used with adequate ventilation. The TLV for phosphoric acid is 1 mg/m³. The PEL is the same. For personal protective measures, refer to section 4.3.1.

If heated to decomposition, phosphoric acid (ortho) emits toxic oxides of phosphorus fumes.

5.16.6 Storage. Phosphoric acid (ortho) must be stored in cool, dry well ventilated place in tightly closed containers. Refer to section 4.3.2.

5.16.7 Disposal. For appropriate procedures, contact the Installation Environmental Office, the DRMO, or Safety and Health Offices. Refer to section 4.4.

Phosphoric acid has an EPA Hazardous Waste Classification - Corrosive, Waste Number D002.

5.17 Name. PHOSPHOROUS ACID (HYPO), (50%), ANALYZED REAGENT.

H₃PO₂ FW 66.00

Phosphinic acid

5.17.1 Technical description. Phosphorous acid (hypo)(pure) exist as colorless oily liquid or deliquescent crystals with a density of 1.493 g/cm³ at 19°C. Its melting point is 26.5°C and decomposes at 130°C. It is soluble in cold water and very soluble in hot water, alcohol and ether.

5.17.2 Specification. Federal, O-C-265, Chemicals, Analytical; General Specification For.

5.17.3 Requirements. The Federal Specification requirements for phosphorous acid (hypo), 50% analyzed reagent is shown in Table XVIII.

TABLE XVIII. Phosphorous acid (hypo), 50%, analyzed reagent - requirements (USP).

Assay, (H ₃ PO ₂), min, % by wt	48.0
Appearance	A colorless to faintly yellow liquid
Maximum Limits of Impurities (% by wt)	
Chloride (Cl), white, insoluble residue remaining	Negligible
Phosphate (PO ₄), Precipitate formed	Slight
Sulfate (SO ₄) max, (mg)	0.2

5.17.4 Use. Phosphorous acid (hypo), 50% analyzed reagent is intended for use as a reagent in analytical chemistry. It is also used in medicine.

MIL-STD-1443

5.17.5 Safety. Phosphorous acid (hypo), is corrosive and a strong reducing agent. It is irritating to the eyes, skin and mucous membranes. Inhalation of vapors, dust or mist must be avoided. Contact with the eyes, skin and clothing must be avoided. Phosphorous acid (hypo) must be used with adequate ventilation. For personal protective measures, refer to section 4.3.1.

It is a dangerous fire and explosive risk in contact with oxidizing material. If heated to decomposition it emits toxic oxides of phosphorous fumes.

5.17.6 Storage. Phosphorous acid (hypo) must be stored in cool, dry well ventilate place in tightly closed acid resistant containers away from sources of heat and oxidizing materials. Refer to section 4.3.2.

5.17.7 Disposal. For appropriate procedures, contact the Installation Environmental Office, the DRMO, or Safety and Health Offices. Refer to section 4.4.

Phosphorous acid (hypo) has an EPA Hazardous Waste Classification - Corrosive, Waste Number D002.

5.18 Name. PHOSPHOROUS ACID, ANALYZED REAGENT H_3PO_3 FW 82.00
Orthophosphorous acid

5.18.1 Technical description. Phosphorous acid (pure) exists as colorless to yellow deliquescent crystals with a specific gravity of 1.651 at 21°/4°C. Its melting point is 73.6°C and decomposes at 200°C. Its solubility in water is 309 g/100cm³ at 0°C and 649 g/100cm³ at 40°C. It is soluble in alcohol. It absorbs oxygen very readily with the formation of orthophosphoric acid (H_3PO_4).

5.18.2 Specification. Military, MIL-P-13981, Phosphorous Acid, Analyzed Reagent.

5.18.3 Requirements. The Military Specification requirements for phosphorous acid, analyzed reagent is shown in Table XIX.

TABLE XIX. Phosphorous acid, analyzed reagent - requirements.

Assay (as H_3PO_3), min, % by wt	98.00
Maximum Limits of Impurities (% by wt)	
Insoluble matter	0.005
Chloride (Cl)	0.003
Phosphate (PO_4)	0.030
Sulfate (SO_4)	0.002
Heavy metals (as Pb)	0.002

5.18.4 Use. Phosphorous acid, analyzed reagent is intended for use as a reagent in analytical chemistry.

5.18.5 Safety. Phosphorous acid is a corrosive and strong reducing agent. It is irritating to the eye, skin and mucous membranes. Inhalation of vapors, dust or mist must be avoided. Contact with the eyes, skin and

MIL-STD-1443

clothing must be avoided. Phosphorous acid must be used with adequate ventilation. For personal protective measures, refer to section 4.3.1.

Phosphorous acid is a dangerous fire and explosive risk in contact with oxidizing materials. If heated to decomposition it emits toxic oxides of phosphorous fumes.

5.18.6 Storage. Phosphorous acid must be stored in a cool, well ventilated place in tightly closed acid resistant containers away from sources of heat and oxidizing materials. Refer to section 4.3.2.

5.18.7 Disposal. For appropriate procedures, contact the Installation Environmental Office, the DRMO, or Safety and Health Offices. Refer to section 4.4.

Phosphorous acid has an EPA Hazardous Waste Classification - Corrosive, Waste Number D002.

5.19 Name. PHOSPHOROUS PENTOXIDE, ANALYZED REAGENT. P_2O_5 FW 141.94
 Phosphoric anhydride
 Diphosphorus pentoxide
 Phosphorus (V) oxide

5.19.1 Technical description. Phosphorous pentoxide (pure) exists as white, deliquescent monoclinic crystals or powder with a density of 2.3 g/cm³. It has a melting point of 580-585°C and sublimates at 300°C. It decomposes in cold water to H_3PO_4 . It is soluble in H_2SO_4 and insoluble in acetone.

5.19.2 Specification. Federal, O-C-265, Chemicals, Analytical; General Specification For.

5.19.3 Requirements. The Federal Specification requirements for phosphorous pentoxide, analyzed reagent is shown in Table XX.

TABLE XX. Phosphorus pentoxide, analyzed reagent - requirements (ACS).

Assay (P_2O_5), min, (% by wt)	98.0
Maximum Limits of Impurities (% by wt)	
Insoluble matter	0.02
Phosphorus trioxide (P_2O_3) to pass color test	Limit about 0.02
Ammonium (NH_4)	0.01
Heavy metals (as Pb)	0.01

5.19.4 Use. Phosphorus pentoxide, analyzed reagent is intended for use as a reagent in analytical chemistry. It is also used in pharmaceuticals and organic synthesis.

5.19.5 Safety. Phosphorus pentoxide is corrosive and nonflammable. It is irritating to the eyes, skin and mucous membranes. Inhalation of dust and mist must be avoided. Contact with the eyes, skin and clothing must be avoided. Phosphorus pentoxide must be used with adequate ventilation. For personal protective measures, refer to section 4.3.1.

MIL-STD-1443

Phosphorus pentoxide reacts violently with water, evolving heat and acrid white fumes, forming phosphoric acid. The reaction with alcohol is similar. The TLV for phosphoric acid is 1 mg/m³. The PEL is the same.

5.19.6 Storage. Phosphorus pentoxide must be stored in cool, dry, well ventilated place in tightly closed containers. Refer to section 4.3.2.

5.19.7 Disposal. For appropriate procedures, contact the Installation Environmental Office, the DRMO, or Safety and Health Offices. Refer to section 4.4.

Phosphorus pentoxide has an EPA Hazardous Waste Classification - Corrosive - Waste Number D002; Reactive - Waster Number D003.

5.20 Name. SELENIOUS ACID, ANALYZED REAGENT. H₂SeO₃ FW 128.97
 Monohydrated selenium dioxide
 Selenous acid

5.20.1 Technical description. Selenious acid (pure) exists as colorless, hexagonal deliquescent crystals with a density of 3.004 g/cm³ at 15°/4°C. It gives off water upon heating and decomposes at 70°C. It has a solubility of 167 g/100cm³ water at 20°C and is very soluble in hot water. It is very soluble in alcohol and insoluble in ammonia. It is oxidized to selenic acid by strong oxidizing agents such as ozone, hydrogen dioxide and chlorine. It is reduced to selenium by most reducing agents including hydriodic acid, sulfurous acid, sodium hyposulfite and hypophosphorous acid.

5.20.2 Specification. Federal, O-C-265, Chemicals, Analytical; General Specification For.

5.20.3 Requirements. The Federal Specification requirements for selenious acid, analyzed reagent is shown in Table XXI.

TABLE XXI. Selenious acid, analyzed reagent - requirements (USP).

Appearance	colorless or white crystals
Assay (H ₂ SeO ₃), min, % by wt	93.0
Maximum Limits of Impurities (% by wt)	
Insoluble matter	The solution is clear and complete
Residue on ignition (% by wt)	0.01
Selenate and sulfate, turbidity or precipitate formed	none

5.20.4 Use. Selenious acid, analyzed reagent is intended for use as a reagent in analytical chemistry for determination of alkaloids.

5.20.5 Safety. Selenious acid is high toxic by inhalation and moderately toxic by skin absorption. Inhalation of selenium compounds causes poisoning via rapid gut and lung absorption. It is most damaging to liver, kidneys, spleen, bone marrow and thyroid. It is irritating to the eyes, skin and mucous membranes. Inhalation of dust and mist must be avoided. Contact with the eyes, skin and clothing must be avoided. Selenious acid

MIL-STD-1443

must be used with adequate ventilation. Selenious is a suspected carcinogen. The TLV for selenium compounds (as Se) is 0.2 mg/m³. The PEL is the same. For personal protective measures, refer to section 4.3.1.

5.20.6 Storage. Selenious acid must be stored in cool, dry, well ventilated place in tightly closed containers. Refer to section 4.3.2.

5.20.7 Disposal. For appropriate procedures, contact the Installation Environmental Office, the DRMO, or Safety and Health Offices. Refer to section 4.4.

Selenious acid has an EPA Hazardous Waste Classification - Toxic, Waste Number U204.

5.21 Name. SULFAMIC ACID, ANALYZED REAGENT. NH₂SO₃H FW 97.09
Amidosulfonic acid
Aminosulfonic acid

5.21.1 Technical description. Sulfamic acid (pure) exists as colorless rhombic crystals with a density of 2.126 g/cm³ at 25°C. Its melting point is 200°C with decomposition. It also decomposes in hot water. Its solubility in cold water is 14.68 g/cm³ and 47.08 g/cm³ at 80°C. It is very slightly soluble in alcohol, ether and acetone. It is insoluble in carbon-tetrachloride.

5.21.2 Specification. Federal, O-C-265, Chemicals, Analytical; General Specification For.

5.21.3 Requirements. The Federal Specification requirements for sulfamic acid, analyzed reagent is shown in Table XXII.

TABLE XXII. Sulfamic acid, analyzed reagent - requirements (USP).

Appearance	colorless or white crystals
Assay (NH ₂ SO ₃ H), min, % by wt	98.5
Maximum Limits of Impurities (% by wt)	
Insoluble matter	0.01
Residue on ignition	0.01
Chloride (Cl)	0.001
Heavy metal (as Pb)	0.001
Iron (Fe)	0.0005
Sulfate (SO ₄)	0.05

5.21.4 Use. Sulfamic acid, analyzed reagent is intended for use as a reagent in analytical chemistry including use as a standard in alkalimetry.

5.21.5 Safety. Sulfamic acid is a corrosive material and nonflammable. Its water solutions are strongly acidic. It is irritating to the eyes, skin, respiratory tract and mucous membranes. Inhalation of dust and mist must be avoided. Contact with the eyes, skin and clothing must be avoided. Sulfamic acid must be used with adequate ventilation. For personal protective measures, refer to section 4.3.1.

MIL-STD-1443

Sulfamic acid is stable when dry, but in solution it slowly hydrolyzes forming ammonium bisulfate. Hazardous gases can be evolved; e.g., flammable hydrogen from contact with some metals, poisonous hydrogen cyanide from cyanide salts and poisonous hydrogen sulfide from sulfides. If heated to decomposition, sulfamic acid may emit toxic sulfur dioxide, sulfur trioxide and ammonia.

5.21.6 Storage. Sulfamic acid must be stored in a cool, dry, well ventilated place in tightly closed containers away from sources of heat and open flame and reducing materials. Protect from light. Refer to section 4.3.2.

5.21.7 Disposal. For appropriate procedures, contact the Installation Environmental Office, the DRMO, or Safety and Health Offices. Refer to section 4.4.

Sulfamic acid has an EPA Hazardous Waste Classification - Corrosive, Waste Number D002.

5.22 Name. SULFURIC ACID, ANALYZED REAGENT. H_2SO_4 FW 98.08
Oil of vitriol

5.22.1 Technical description. Sulfuric acid (pure) exists as a clear, oily liquid with a specific gravity of 1.841 at 20°/4°C for 96-98% material. Its melting point is 10.36°C for 100% material. It is soluble in all proportions in cold water with the evolution of heat and soluble in all proportions in hot water. It decomposes when heated with a loss of SO_3 until a concentration of 98.33% is reached which boils at 338°C without further change in concentration. Sulfuric acid is a diprotic acid which ionizes in two stages. In dilute solution, primary ionization is complete, while the secondary ionization is less complete. It can form both normal and hydrogen sulfate salts. Hot concentrated sulfuric acid can act as an oxidizing agent depending on the strength of the reducing agent with which it acts.

5.22.2 Specification. Federal, O-C-265, Chemicals, Analytical; General Specification For.

5.22.3 Requirements. The Federal Specification requirements for sulfuric acid, analyzed reagent is shown in Table XXIII.

TABLE XXIII. Sulfuric acid, analyzed reagent - requirements (ACS).

Appearance	Free from suspended or insoluble matter as received and after dilution to 2N.
Assay (H_2SO_4), min-max, % by wt	95.0-98.0
Color (APHA), as received and after dilution to 2N, max	10
Maximum Limits of Impurities (ppm)	
Residue after ignition	5
Chloride (Cl)	0.2
Nitrate (NO_3)	0.5
Ammonium (NH_4)	2

MIL-STD-1443

TABLE XXIII. Sulfuric acid, analyzed reagent - requirements (ACS)
- Continued.

Substances reducing permanganate, to pass color test (as SO ₂)	2
Arsenic (As)	0.01
Heavy metals (as Pb)	1
Iron (Fe)	0.2
Maximum Limits of Impurities (ppb)	
Mercury (Hg)	5

5.22.4 Use. Sulfuric acid, analyzed reagent is intended for use as a reagent in analytical chemistry. It is also used in metallurgy and as a alkylation catalyst.

5.22.5 Safety. Sulfuric acid is nonflammable , very reactive and corrosive. Contact with the body results in rapid destruction of tissue, causing severe burns. It is irritating to the eyes, skin and mucous membranes. Inhalation of vapors and mist must be avoided. Contact with the eyes, skin and clothing must be avoided. These are systemic effects secondary to tissue damage caused by contact. Repeated contact with dilute solutions can cause a dermatitis, and repeated or prolonged inhalation of the mist can cause an inflammation of the upper respiratory tract. Inhalation of concentrated vapor or mist from hot acid can cause rapid loss of consciousness with serious damage to lung tissue. Sulfuric acid must be used with adequate ventilation. The TLV for sulfuric acid is 1 mg/m³. The PEL is the same. For personal protective measures, refer to section 4.3.1.

Sulfuric acid, as an oxidizer, can cause ignition of combustibles. It can react with metals to emit flammable hydrogen gas. If heated it can emit toxic fumes.

5.22.6 Storage. Sulfuric acid must be stored in cool, dry, well ventilated place in tightly closed acid resistant containers away from reducing and combustible materials. Protect containers against physical damage. Refer to section 4.3.2.

5.22.7 Disposal. For appropriate procedures, contact the Installation Environmental Office, the DRMO, or Safety and Health Offices. Refer to section 4.4.

Sulfuric acid has an EPA Hazardous Waste Classification - Corrosive, Waste Number D002; Reactive, Waste Number D003.

5.23 Name. SULFURIC ACID, 18N, , ANALYZED REAGENT. H₂SO₄ FW 98.08
Oil of vitriol

5.23.1 Technical description. Sulfuric acid (pure) exists as a clear, colorless, oily liquid with a specific gravity of 1.841 at 20°/4°C for 96-98% material. Its melting point is 10.36°C for 100% material. It is soluble in all proportions in cold water with the evolution of heat and soluble in all proportion in hot water. It decomposes in alcohol. Sulfuric acid decomposes when heated with a loss of SO₃ until a concentration of 98.33% is reached which boils at 338°C without further change in concen-

MIL-STD-1443

tration. Sulfuric acid is a diprotic acid which ionizes in two stages. In dilute solution, primary ionization is complete, while the secondary ionization is less complete. It can form normal and hydrogen sulfate salts. Sulfuric acid can act as an oxidizing agent. Sulfuric acid, 18N, has a specific gravity of 1.5 at 20°/4°C.

5.23.2 Specification. Meets requirements for sulfuric acid, 18N, analyzed reagent used in Standard Test Method, ASTM 874.

5.23.3 Requirements. Sulfuric acid, 18N, analyzed reagent is commercially available with requirements as shown in Table XXIV.

TABLE XXIV. Sulfuric acid, 18N, analyzed reagent - requirements.

Normality	17.5-18.5
Maximum Limits of Impurities (% by wt)	
Ammonium	0.0002
Arsenic	0.000001
Chloride	0.00002
Heavy metals	0.0001
Iron	0.00002
Nitrate	0.00005
Residue after ignition	0.0005
Substances reducing permanganate as SO ₂ , to pass test	Limit about 0.0002

5.23.4 Use. Sulfuric acid, 18N, analyzed reagent is intended for use as a reagent for the determination of sulfated ash from lubricating oils and additives using the Standard Test Method, ASTM 874. It is also used where sulfuric acid must be added to aqueous media or diluted to make standard solutions because it is safer to handle than concentrated sulfuric acid.

5.23.5 Safety. Sulfuric acid is nonflammable, reactive and corrosive. Contact with the body results in rapid destruction of tissue, causing severe burns. It is irritating to the eyes, skin and mucous membranes. Inhalation of vapors and mist must be avoided. Contact with the eyes, skin and clothing must be avoided. There are systemic effects secondary to tissue damage caused by contact. Repeated contact with dilute solutions can cause a dermatitis, and repeated or prolonged inhalation of the mist can cause inflammation of the upper respiratory tract. Inhalation of concentrated vapor or mist from hot acid can cause rapid loss of consciousness with serious damage to lung tissue. Sulfuric acid must be used with adequate ventilation. The TLV for sulfuric acid is 1 mg/m³. The PEL is the same. For personal protective measures, refer to section 4.3.1.

Sulfuric acid, as an oxidizer, can cause ignition of combustibles. It can react with metals to emit flammable hydrogen gas. If heated it can emit toxic fumes.

5.23.6 Storage. Sulfuric acid must be stored in a cool, dry, well ventilated place in tightly closed, acid resistant containers away from reducing and combustible materials. Protect container against physical damage. Refer to section 4.3.2.

MIL-STD-1443

5.23.7 Disposal. For appropriate procedures, contact the Installation Environmental Office, the DRMO, or Safety and Health Offices. Refer to section 4.4.

Sulfuric acid has EPA Hazardous Waste Classification - Corrosive, Waste Number D002; Reactive, Waste Number D003.

5.24 Name. SULFURIC ACID, FUMING, ANALYZED REAGENT. $H_2S_2O_7$ FW 178.14
Oleum
Pyrosulfuric acid

5.24.1 Technical description. Fuming sulfuric acid containing one mole of SO_3 /mole H_2SO_4 (pure) exists as colorless hygroscopic crystals with a density of 1.9 g/cm^3 at 20°C . Its melting point is 35°C . It decomposes in cold and hot water and alcohol.

5.24.2 Specification. Federal, O-C-265, Chemicals, Analytical; General Specification For.

NOTE: This specification applies to fuming sulfuric acid with nominal content of 15, 20, or 30% free SO_3 .

5.24.3 Requirements. The Federal Specification requirements for sulfuric acid, fuming, analyzed reagent is shown in Table XXV.

TABLE XXV. Sulfuric acid, fuming, analyzed reagent - requirements (ACS).

Appearance	colorless to very light brown color
Assay (free SO_3), min-max, % by wt	15.0-18.0 20.0-23.0 30.0-33.0
Maximum Limit of Impurities (% by wt)	
Residue after ignition	0.002
Maximum Limit of Impurities (ppm)	
Nitrate (NO_3)	1
Ammonium (NH_4)	3
Arsenic (As)	0.03
Iron (Fe)	2

Sulfuric acid, fuming, analyzed reagent is also commercially available with requirements as shown in Table XXVI.

TABLE XXVI. Sulfuric acid, fuming, analyzed reagent - requirements.

Appearance	to pass test
Assay (free SO_3), min-max, % by wt	12.0-17.0 18.0-24.0 27.0-33.0
Maximum Limit of Impurities (% by wt)	
Residue after ignition	0.002
Maximum Limit of Impurities (ppm)	
Nitrate (NO_3)	0.1
Ammonium (NH_4)	0.3

MIL-STD-1443

TABLE XXVI. Sulfuric acid, fuming, analyzed reagent - requirements - Continued.

Arsenic (As)	0.03
Iron (Fe)	0.2

5.24.4 Use. Sulfuric acid, fuming, analyzed reagent is intended for use as a reagent in analytical chemistry.

5.24.5 Safety. Sulfuric acid, fuming, is corrosive and very reactive. Contact with the body results in rapid destruction of tissue, causing severe burns. It is irritating to the eyes, skin and mucous membranes. Inhalation of vapors and mist must be avoided. Contact with the eyes, skin and clothing must be avoided. There are systemic effects secondary to tissue damage caused by contact. Repeated contact with dilute solutions can cause a dermatitis, and repeated or prolonged inhalation of the mist can cause an inflammation of the upper respiratory tract. Inhalation of concentrated vapor or mist from hot acid can cause rapid loss of consciousness with serious damage to lung tissue. Sulfuric acid, fuming, must be used with adequate ventilation. For personal protective measures, refer to section 4.3.1.

Sulfuric acid, fuming, is dangerous by chemical reaction with reducing agents. If heated to decomposition it emits toxic fumes of oxides of sulphur. It will react with water or steam to produce heat and toxic and corrosive fumes.

5.24.6 Storage. Sulfuric acid, fuming, must be stored in a cool, dry, well ventilated place in tightly closed acid resistant containers away from combustible materials. Protect containers against physical damage. Refer to section 4.3.2.

5.24.7 Disposal. For appropriate procedures, contact the Installation Environmental Office, the DRMO, or Safety and Health Offices. Refer to section 4.4.

Sulfuric acid, fuming, has an EPA Hazardous Waste Classification - Corrosive, Waste Number D002; Reactive, Waste Number D003.

5.25 Name. SULFUROUS ACID, ANALYZED REAGENT. H_2SO_3 FW 82.08
A solution of SO_2 in water

5.25.1 Technical description. Sulfurous acid (pure) exists as colorless solution of sulfur dioxide in water. It has a density of about 1.03. It is soluble in cold water, alcohol, ether and acetic acid. It has a suffocating odor of sulfur dioxide. It gradually oxidizes in air to sulfuric acid.

5.25.2 Specification. Federal, O-C-265, Chemicals Analytical, General Specification For.

5.25.3 Requirements. The Federal Specification requirements for sulfurous acid, analyzed reagent is shown in Table XXVII.

MIL-STD-1443

TABLE XXVII. Sulfurous acid, analyzed reagent - requirements (ACS).

Assay (SO ₂), min, % by wt	6.0
Maximum Limits of Impurities (% by wt)	
Residue after ignition	0.005
Maximum Limits of Impurities (ppm)	
Chloride (Cl)	5
Arsenic (As)	0.5
Heavy metals (as Pb)	0.005
Iron (Fe)	5

5.25.4 Use. Sulfurous acid, analyzed reagent is intended for use as a reagent in analytical chemistry.

5.25.5 Safety. Sulfurous acid, is a corrosive liquid. It is irritating to the eyes, skin and mucous membranes. Inhalation of fumes or mist must be avoided. Contact with the eyes, skin and clothing must be avoided. Sulfurous acid must be used with adequate ventilation. For personal protective measures, refer to section 4.3.1.

If heated to decomposition, sulfurous acid emits highly toxic fumes of SO₂.

5.25.6 Storage. Sulfurous acid must be stored in a cool, dry, well ventilated place in tightly closed containers. Refer to section 4.3.2.

5.25.7 Disposal. For appropriate procedures, contact the Installation Environmental Office, the DRMO, or Safety and Health Offices. Refer to section 4.4.

Sulfurous acid has an EPA Hazardous Waste Classification - Corrosive, Waste Number D002.

5.26 Name. SULFUR TRIOXIDE, STABILIZER, ANALYZED REAGENT
SO₃ FW 80.06
Sulfuric anhydride

5.26.1 Technical description. Sulfur trioxide (pure) exists as a colorless, clear liquid with a density of 1.920 (H₂O=1) at 20°/4°C. Its melting point is 16.8°C and boiling point is 44.8°C. It decomposes in hot and cold water forming sulfuric acid.

5.26.2 Specification. None.

5.26.3 Requirements. Sulfuric trioxide, stabilized, analyzed reagent is commercially available with a typical analysis as shown in Table XXVIII.

TABLE XXVIII. Sulfur trioxide, stabilized, analyzed reagent - typical analysis.

Sulfur trioxide (SO ₃), % by wt	99.6
Sulfur acid (H ₂ SO ₄), % by wt	0.2
Stabilizer, % by wt	0.2
Iron (Fe), ppm	5

MIL-STD-1443

5.26.4 Use. Sulfur trioxide, stabilized reagent is intended for use as a reagent for the sulfonation of organic compounds.

5.26.5 Safety. Sulfur trioxide is highly toxic, a strong oxidizer and corrosive. It is irritating to the eyes, skin and mucous membranes. Inhalation of vapors and mist must be avoided. Contact with the eyes, skin and clothing must be avoided. Sulfur trioxide must be used with adequate ventilation. For personal protective measures, refer to section 4.3.1.

Absolutely dry sulfur trioxide is not corrosive to metals and shows no acid reaction. On exposure to air, it absorbs moisture rapidly, emitting dense white fumes. It combines with water with explosive violence, forming sulfuric acid. Due to its affinity for water sulfur trioxide chars many organic substances. On contact with some materials the heat produced by dehydration is sufficient to cause fire.

5.26.6 Storage. Sulfur trioxide, stabilizer must be stored in cool, dry, well ventilated place in tightly closed containers away from reducing or flammable materials. Protect against physical damage. Refer to section 4.3.2.

5.26.7 Disposal. For appropriate procedures, contact the Installation Environmental Office, the DRMO, or Safety and Health Offices. Refer to section 4.4.

Sulfur trioxide, stabilized has an EPA Hazardous Waste Classification - Reactive, Waste Number D003.

MIL-STD-1443

6. NOTES

6.1 Subject term (key word) listing.

Arsenic pentoxide, analyzed reagent
Arsenic pentoxide, n-hydrate analyzed reagent
Arsenic trioxide, analyzed reagent
Boric acid, analyzed reagent
Chlorosulfonic acid, reagent
Chromium trioxide, analyzed reagent
Exposure limits, hazardous chemicals
Hazardous wastes, disposal and storage of
Hydrobromic acid (48%), analyzed reagent
Hydrochloric acid, analyzed reagent
Hydrofluoric acid, analyzed reagent
Information, hazardous chemicals
Iodic acid, analyzed reagent
Iodic pentoxide, analyzed reagent
Molybdenum trioxide, analyzed reagent
Nitric acid, analyzed reagent
Phosphomolybdic acid, analyzed reagent
Phosphoric acid (meta), analyzed reagent
Phosphoric acid (othro) , analyzed reagent
Phosphorous acid (hypo), 50%, analyzed reagent
Phosphorous acid, analyzed reagent
Phosphorus pentoxide, analyzed reagent
Safety, hazardous chemicals
Selenious acid, analyzed reagent
Sulfamic acid, analyzed reagent
Sulfuric acid, analyzed reagent
Sulfuric acid, 18N, analyzed reagent
Sulfuric acid, fuming, analyzed reagent
Sulfurous acid, analyzed reagent
Sulfur trioxide, stabilized, analyzed reagent

6.2 Abbreviations. The use of abbreviations shall be in accordance with MIL-STD-12 where applicable. Metric system abbreviations and symbols shall be in accordance with ASTM E 380.

MIL-STD-1443

Preparing activity: Army - EA

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Custodians:

Army - EA

Navy - SH

Air Force - 68

Reviewer activities:

Army - AR, AV, MD, MI, SM

Navy - SA

DOD - DM, DP, GS,

User activities:

Army - ME, MI

Navy - MC, OS, YD

DOD - DM

STANDARDIZATION DOCUMENT IMPROVEMENT PROPOSAL

(See Instructions - Reverse Side)

1. DOCUMENT NUMBER

MIL-STD-1443

2. DOCUMENT TITLE

INORGANIC ACIDS AND ACID ANHYDRIDES, REAGENT GRADE

3a. NAME OF SUBMITTING ORGANIZATION

4. TYPE OF ORGANIZATION (Mark one)

☐

VENDOR

☐

USER

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OTHER (Specify): _____

b. ADDRESS (Street, City, State, ZIP Code)

5. PROBLEM AREAS

a. Paragraph Number and Wording:

b. Recommended Wording:

c. Reason/Rationale for Recommendation:

6. REMARKS

7a. NAME OF SUBMITTER (Last, First, MI) - Optional

b. WORK TELEPHONE NUMBER (Include Area Code) - Optional

c. MAILING ADDRESS (Street, City, State, ZIP Code) - Optional

8. DATE OF SUBMISSION (YYMMDD)

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